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# BITSAT 2013

# Previous Year Questions

## CHEMISTRY

Single	Single correct answer type					
41.	For the properties mentioned, the correct trend for the different species is in					
Ans.	<ul> <li>(A) Strength as Lewi</li> <li>(B) Inert pair effect –</li> <li>(C) Oxidising propert</li> <li>(D) First ionisaiton et</li> <li>(D)</li> </ul>	$y - A\ell^{3+} > \ell n^{3+} > Ti^{3+}$	SaCℓ₃			
Sol.	As we know on movi	ng down the group first i	onization enthalpy decre	eases top to bottom therefore order		
	of first ionization entl	nalpy for group 13 eleme	ent is			
	$B > A\ell > Ga > In > T$	l				
40	Debr theory is epulie	abla ta				
42.	Bohr theory is applic (A) He	(B) Li <sup>2+</sup>	(C) He <sup>2+</sup>	(D) None of these		
Ans.	(B)					
Sol.		icable to H-like species o	containing one electron	only for e, Li <sup>2+</sup> .		
43.	Using MOT, which o	f the following pair denot	es paramagnetic specie	es		
	(A) $B_2$ and $C_2$	(B) $B_2$ and $O_2$	(C) $N_2$ and $C_2$	(D) $O_2$ and $O_2^{2-}$		
Ans.	(B)			2		
Sol.	Among given four pairs, $B_2$ and $O_2$ are paramagnetic due to presence of unpaired electron.					
	MO(EC) of B <sub>2</sub> = $\sigma 1s^2 \sigma^* 1s^2 \sigma 2s^2 \sigma^* 2s^2 \pi 2p_2^1 \equiv \pi 2p_y^1$					
44.	0.41g of metal combines with 46.6 mL of oxygen at STP. The equivalent weight of metal is					
	(A) 12	(B) 24	(C) 18	(D) 36		
Ans.	(A)					
Sol.	1 mole of O2 = 4 eq. of Of oxygen					
	$22400 \mathrm{m}\ell$ of $O_2 = \frac{4}{22400} \times 46.6$					
	= 0.00832 eq.					
	Equivalent of metal = Equivalent of oxygen					
	$\frac{\text{Weight}}{\text{Equivalent}} = 0.00832$					
	$\frac{0.1}{E} = 0.00832$					
	$\therefore = \frac{0.1}{0.00832}$	= 12.0				
45.	Which of the followin	g choice represent corre	ect order of first ionizatio	on enthalpy		

**45.** Which of the following choice represent correct order of first ionization enthalpy

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(A) B < C < N < O < F (B) B > C > N > 0 > F (C) B < C < N > O < F (D) B < C < N > O > F Ans. (C) Sol. lonization energy is the minimum amount of energy required to remove the outermost electron from an isolated gaseous atom. Quantitatively, it depends on the attraction between electron present on outermost shell and nucleus. Greater the interaction between outermost electron and nucleus, higher will be its ionization enthalpy. So correct order of first it must be B < C < N < O < FBut due to extra stable half-filled electronic configuration of p orbital of N has more value of first ionization enthalpy than oxygen hence, correct order is B < C < N > O > F46. Which of the following reaction produces most stable alkene? (A) 2-chloro butane (B) 2, 3-dichloro butane (C) 2, 2-dichloro butane (D) 2, 3-dichloro, 2, 3-dimethyl butane Ans. (D) OUMDATIC Sol. Molecular structure of given names of organic compounds are written as 2 -chloro butane 2, 3-dichlorobutane C CH<sub>3</sub> CH<sub>3</sub> CH<sub>3</sub> 2, 2-dichlorobutane CH<sub>3</sub> 2, 3-dichloro-2,3-dimethy1 butane According to Saytzeff's rule, more substituted (alkylated) alkene are more stable. When the alkyl halide is treated with base, it undergo elimination reaction and produces alkane as follows: CH<sub>3</sub> Cľ (a) (b) (d) (c) ĊI Ċ Base Base Base Base

<b>Conjugation</b> Greater the conjugation greater will be the stability of product. Hence, D has maximum stability, the correct choice is (2, 3-dichloro, 2, 3-dimethyl butane) which is tabilised by conjugation as well as Saytzeff's rule.

47. Which of the following is less acidic among the given halogen compounds?

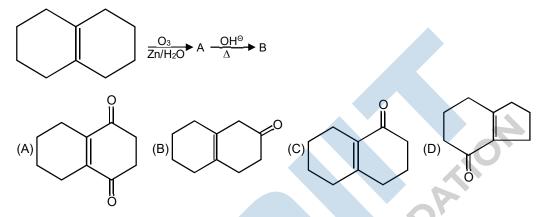
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Α

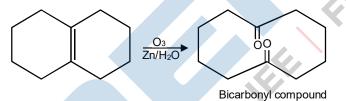
(A) CHF<sub>3</sub> (B) CH $\ell_3$  (C) CHC $\ell_3$  (D) CHBr<sub>3</sub>

Ans. (B)

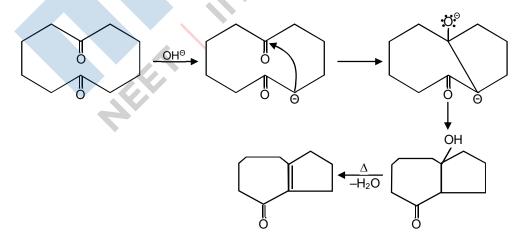
- **Sol.** Due to stronger –I effect of F than CI, CHF<sub>3</sub> should be more acidic than CHCL<sub>3</sub>. But actually reverse is true. This is due to :  $CC\ell_3^-$  left after the removal of a proton from CHCL<sub>3</sub> is stabilized by resonance due to presence of d-orbitals in CI than :  $CF_3^-$  left after the removal of a proton from CHF<sub>3</sub> which is not stabilized by resonance due to the absence of d-orbitals on F.
- 48. What will be the final product of the reaction?



- Ans. (D)
- **Sol.** Ozonolysis on ozonolysis the given alken undergo ozonide formation followed by reduction to produce bucarbonyl compound as



Now, this bicarbonyl compound undergoes interamolecular aldol condensation as follows:



- 49. The vapour pressure of a solvent decreased by 10 mm of Hg when a non-volatile solute was added to the solvent. The mole fraction of solute in solution is 0.2, what would be the mole fraction of solvent if the decrease in vapour pressure is 20 mm of Hg
  (A) 0.8 (B) 0.6 (C) 0.4 (D) 0.3
- Ans. (B)
- **Sol.** This question is based on Raoult's law. It represent that the partial pressure of each component in the solution is directly proportional to its mole fraction for a solution i.e.,  $P_A \propto \chi_A$  and  $P_B \propto \chi_B$ . From Raoult's law

 $p^{O} - p_{s} = p^{O} \times Mole$  fraction of solute 10 =  $p^{O} \times 0.2$ 

 $20 = p^{O} \times \chi_{2}$ 

 $\therefore \chi_2 = 0.4$ 

And  $\chi_1 = 1 - 0.4 = 0.6$ 

 $\chi_1$  = mole fraction of solvent

**50.** Chose the law that corresponds to data shown for the following reaction,  $A + B \rightarrow$  products

Exp.	[A]	[B]	Initial rate
1	0.012	0.035	0.1
2	0.024	0.070	0.8
3	0.024	0.035	0.1
4	0.012	0.070	0.8
(A) Rate = k[B]	(B) Rate = k[B] <sup>4</sup>		

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(C) Rate = k[A] [B]<sub>3</sub>
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Ans. (A)
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Sol. It is seen that, in experiments (3) and 2i). [A] is constant and [B] is doubled and rates becomes 8 times, so order w.r.t, [B] = 3 In experiments (1) and (3) [B] is constant and [A] is doubled, but rate does not change, so order w.r.t. [A] = 0 Thus, rate = k[B]<sup>3</sup>

(D) Rate =  $k[A]^{3}[B]$ 

- **51.** The magnitude of  $\Delta_0$  will be highest in which (A) [Cr (CN<sub>6</sub>)]<sup>3-</sup> (B) [Cr (H<sub>2</sub>O)<sub>6</sub>]<sup>3+</sup> (C) [Cr (NH<sub>3</sub>)<sub>6</sub>]<sup>3+</sup> (D) [Cr (C<sub>2</sub>O<sub>4</sub>)<sub>3</sub>]<sup>3-</sup>
- Ans. (A)
- **Sol.** The crystal field splitting,  $\Delta_0$  depends upon the field produced by the ligand and change on the metal ion. In all these complexes of chromium, charge acquired by metal ion is +3. Therefore  $\Delta_0$  depends upon

the field produced by the ligand. In accordance with the spectrochemical series, the increasing order of field strength is  $C_2O_4^{2-} < H_2O < NH_3 < CN^-$ 

Thus,  $CN^{-}$  is the strong field ligand and will produce highest magnitude of  $\Delta_0$ .

52. Arrange these in correct order of decreasing reactivity.

$$F \xrightarrow{CH_3} F \xrightarrow{CH_3} OH F \xrightarrow{OH} OH H_3 \xrightarrow{OH} OH H_3 \xrightarrow{OH} OH H_1 \xrightarrow{OH} OH H_1 \xrightarrow{III} IV$$

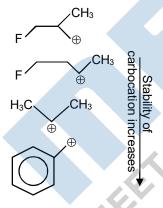
(A) | > || > ||| > |V|(B) I > III > II > IV (C) IV > III > II > I

(D) IV > III > I > II

Ans. (C)

- This problem includes conceptual mixing of carbocation stability and reactivity of alcohol. Sol.
  - Remove the OH group by dehydration and then arrange the carbocation in increasing order of correctly.
  - Order of carbocation stability is same as S<sub>N</sub>1 reactivity of alkyl halides.

<b>Carbocation stability</b> The carbocation is formed during reaction of alcohol by removal of OH. More stable the carbocation more will be its reactivity of carbocation formed during reaction are as follows Traffic



Benzyl carbocation is more stable due to conjugation with phenyl ring.

- 53. When 2-methyl propan-1-ol is treated with a mixture of conc. HC $\ell$  and ZnC $\ell_2$ , turbidity appears immediately due to the formation of
  - (A) 2-methyl propane
  - (C) 2-methyl-2-chloropropane
- (B) 2-methyl propene
- (D) 2-chlorobutane

Ans. (C)

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- **Sol.** When 2-methyl propan-1-ol is treated with a mixture of conc.  $HC\ell$  and  $ZnC\ell_2$  (Lucas reagent) then tertalkyl halide is formed and produced turbidity due to its less solubility.
- 54. Gastric juice in human stomach has pH value about 1.8 and pH of small intestine is about 7.8. The pK<sub>a</sub> value of aspirin is 3.5. Aspirin will be
  - (A) Ionised in the small intestine and stomach
  - (B) Ionised in the stomach and almost unionized in the small intestine
  - (C) Unionised in small intestine and stomach
  - (D) Completely ionized in small intestine and stomach
- Ans. (A)
- **Sol.** Aspirin is a moderate acid (pK<sub>a</sub> = 3.5) Therefore, it is almost unionized in stomach due to its strong acidic medium. It happens due to common ion effect. On the other hand, in small intestine, the medium is alkaline, hence, aspirin will be sufficiently ionized in it.
- 55. When a solution of potassium chromate is treated with an excess of dilute nitric acid
  - (A) Cr3+ and  $Cr_2O_7^{2-}$  are formed (B)  $Cr_2O_7^{2-}$  and  $H_2O$  are formed
  - (C)  $\operatorname{Cr}_2 O_4^{2-}$  reduced to  $\operatorname{Cr}^{3+}$  (D)  $\operatorname{Cr}O_4^{2-}$  oxidized to  $\operatorname{Cr}_2 O_7^{2-}$  only
- Ans. (B)
- **Sol.** The reaction of  $K_2CrO_4$  with dilute nitric acid is represent as  $2CrO_4^{2-} + 2H^+ \rightarrow Cr_2O_7^{2-} + H_2O$
- 56.Calcium carbide reacts with heavy water to form<br/> $(A) C_2D_2$ (B) CaD2(C) CaD2O(D) CD2
- Ans. (A)
- **Sol.** CaC<sub>2</sub> when reacts with water molecule to form acetylene. Similarly, it reacts with D<sub>2</sub>O to form C<sub>2</sub>D<sub>2</sub> CaC<sub>2</sub> + 2D<sub>2</sub>O  $\rightarrow$  C<sub>2</sub>D<sub>2</sub> + Ca (OD)<sub>2</sub>
- 57. Fluorine acts as strongest oxidizing agent because of its high

  (A) Electron affinity
  (B) Ionisation enthalpy
  (C) Hydration enthalpy
  (D) Bond enthalpy

  Ans. (C)
- Sol. Fluorine acts as strongest oxidizing agent due to
  (a) low enthalpy of dissociation of F F bond
  (b) high hydration enthalpy of F<sup>-</sup>
- **58.** The reaction of  $P_4$  with X leads selectively to  $P_4O_6$ . The X is

	(A) dry O <sub>2</sub>	(B) moist O <sub>2</sub>	
	(C) mixture of $O_2$ and $N_2$	(D) O₂ in presence of aquece NaOH	
Ans.	(C)		
Sol.	The reaction of $P_4$ with X leads selectively to $P_4$	$O_6$ because $N_2$ prevents the further reaction of $P_4O_6$ into	
	P <sub>4</sub> O <sub>10</sub> .		
59.	The acidic strength for the hydrides of group 15	follows the order	
	(A) $NH_3 > PH_3 > AsH_3 > sbH_3$	(B) $NH_3 < PH_3 < AsH_3 < sbH_3$	
	(C) $NH_3 > PH_3 > sbH_3 > AsH_3$	(D) $NH_3 < PH_3 < sbH_3 < AsH_3$	
Ans.	(B)		
Sol.	The acidic strength of hydrides is inversely pro	portional to their stability. Since the stability of hydrides	
	decreases from N to Sb. Therefore, the acidic so would be	trength increases from N to Sb. Hence, the correct order	
	$NH_3 < PH_3 < AsH_3 < sbH_3$		
	<b>Caution point</b> As the stability decreases from $NH_3$ to, $BiH_3$ , the reducing character of hydrides		
	increases.		
60.	Which of the following statements are incorrect	in context of borax?	
	(A) It is made up to two triangular BO₃ units and	d two tetrahedral BO4 units	
	(B) One mole of borax can be used as buffer		
	(C) It is a useful primary standard for titrating a	gainst acids	
	(D) Aqueous solution of borax can be used as b	buffer	
Ans.	(B)		
Sol.		of water among 10 molecules forms a part of structure	
	and exists as $Na_2[B_4O_5(OH)_4]8H_2O$		
	[ он ]		
	$Na_2[B_4O_5(OH)_4].8H_2O + 2HCI \rightarrow 2NaCI + 4H_3B$	O <sub>3</sub> + 5H <sub>2</sub> O	
	Methyl orange with pH value of 3.7 is used to de	etect end point. Aqueous solution of borax acts as buffer,	
	because borax is salt is strong base NaOH and	weak acid H <sub>3</sub> BO <sub>3</sub> .	

**61.** Salt  $A + S \rightarrow B \xrightarrow{BaCl_2}$  white precipitate A is paramagnetic in nature and contains about 55% K. Thus,

	A is			
	(A) K <sub>2</sub> O	(B) K <sub>2</sub> O <sub>2</sub>	(C) KO <sub>2</sub>	(D) K <sub>2</sub> SO <sub>4</sub>
Ans.	(C)			

Sol.	Among the given oxides, only KO <sub>2</sub> i.e., potassium superoxide is paramagnetic in nature. This is because
	peroxide ion, $O_2^-$ has three electron bond which makes it paramagnetic and coloured.
	Hence, A is KO <sub>2</sub>

- **62.** When equal volume each of two sols of Agl, one obtained by adding AgNO<sub>3</sub> to slight excess of KI and another obtained by adding KI to slight excess of AgNO<sub>3</sub> are mixed together. It is observed that
  - (A) The sol particles acquired more electric charge
  - (B) The sols coagulated each other mutually
  - (C) A true solution is obtained
  - (D) The two sols stabilized each other

Ans. (B)

- **Sol.** The two sols prepared contains not only Agl but also KI and AgNO<sub>3</sub> as these are taken in excess amounts. When these sols are mixed, the sols being oppositely charged coagulates each other.
- **63.** In the extraction of Ag, Zn is removed from (Zn Ag) alloy through
  - (A) Cupellation

(B) Fractional crystallization

(C) Distillation (D) Electrolytic refining

- Sol. The extraction of Ag using (Zn Ag) alloy is called Parke's process.
   As zinc is volatile at 920° while Ag is not. Thus, on heating (Zn + Ag) alloy, zinc vapourises while Ag remains at the bottom of the vessel. Hence, Zn is removed from (Zn-Ag) alloy through distillation.
- 64. A reaction takes place in three steps. The rate constants are  $k_1$ ,  $K_2$  and  $K_3$ . The overall rate constant  $k = \frac{K_1K_3}{K_2}$ . If  $E_1, E_2$  and  $E_3$  (energy of activation) are 60, 30 and 10kJ respectively, the overall energy of activation is (A) 40 (B) 30 (C) 400 (D) 300
- Ans. (A)
- $\textbf{Sol.} \qquad \textbf{K}_1 \ = \ \textbf{a} e^{\textbf{E}\textbf{a}_1/\textbf{R}\textbf{T}} \ ; \ \textbf{K}_2 \ = \ \textbf{A} e^{\ -\textbf{E}_2/\textbf{R}\textbf{T}}$

$$K_2 = Ae^{-E_3/RT}$$

Overall rate =  $k = \frac{K_1K_3}{K_2}$ 

Ans. (C)

Therefore, overall  $E_a = E_{a_1} + E_{a_3} - E_{a_2}$ = 60 + 10 - 30 = 40kJ

If  $E^{0}_{Fe^{3+}/Fe}$  and  $E^{0}_{Fe^{2+}/Fe}$  are –0.36V and –0.439 respectively, then the value of  $E^{0}_{Fe^{3+}/Fe^{2+}}$  is 65.

(A) (-0.036 -0.439) V	(B) [3(-0.36) + 2(-0.439)] V
(C) (-0.36 - 0.439) V	(D) [3(-0.36) -2(-0.439)] V

Ans. (D)

Given that,  $E_{Fe^{3+}/Fe}^{0} = -0.36 \text{ V};$ Sol.

- $E^0_{Fe^{2^+}/Fe} = -0.439 \, V$
- $Fe^{3+} + 3e^{-} \rightarrow Fe;$  $\Delta G_1 = -3F(E^0_{Fe^{3+}/Fe})$  $\Delta G_2 = -2F(E^0_{Fe^{2^+}/Fe})$  $Fe^{2+} + 2e^{-} \rightarrow Fe;$

$$\mathsf{F} \mathsf{e}^{3_+} + \mathsf{e}^- \to \mathsf{F} \mathsf{e}^{2_+}; \qquad \qquad \Delta \mathsf{G}_3 = - \,\mathsf{F} \Big(\mathsf{E}^0_{\mathsf{F} \mathsf{e}^{3_+}/\mathsf{F} \mathsf{e}}\Big)$$

- $E^{0}_{Fe^{3+}/Fe^{2+}} = 3E^{0}_{Fe^{3+}/Fe} 2E^{0}_{Fe^{2+}/Fe}$ = [3(-0.36) -2(-0.439)] V
- KC  $\ell$  crystallises in the same type of lattice as does NaC  $\ell.$  Given that  $r_{C\ell^-}=0.55$  and 66.

 $r_{K^+} / r_{C\ell^- \Box} = 0.74$ . Determine the ratio of the side of the unit cell for KC $\ell$  to that of NaC $\ell$ 

(A) 0.124	(B) 1.123	(C) 0.891	(D) 1.414
(B)			
Given that,			
$r_{_{Na^{^+}}}/r_{_{C\ell}}=0.55$			

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- Ans. (B)
- Given that, Sol.

 $r_{K^+} / r_{C\ell^-} = 0.75$ 

 $\frac{\mathsf{rKC}\ell}{\mathsf{rNac}\ell}= \texttt{?}$ 

$$r. \frac{r_{Na^+}}{r_{a^-}} = 0.55$$

 $\therefore \frac{I_{Na^+}}{r_{c\ell^-}} + 1 = 0.55 + 1$ 

$$\frac{r_{Na^+} + r_{CI^-}}{r_{CI^-}} = 1.55$$
 .....(i)

$$\therefore \frac{r_{\kappa^{+}}}{r_{c\ell^{-}}} = 0.74$$
$$\therefore \frac{r_{\kappa^{+}}}{r_{c\ell^{-}}} + 1 = 0.74 + 1$$

$$\frac{r_{K^+} + r_{C\ell^-}}{r_{C\ell^-}} = 1.74$$
 .....(ii)

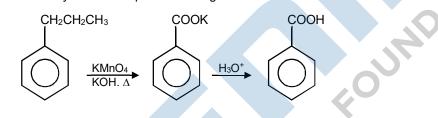
Dividing equations (ii) by (i)

$$\frac{r_{k^+} + r_{C\ell^-}}{r_{Na^+} + r_{c\ell^-}} = \frac{1.74}{1.55} = 1.1226$$

67. The compound formed as a result of oxidation of propyl benzene by KMnO<sub>4</sub> is

(A) Benzaldehye	(B) Benzyl alcohol

- (C) Benzoic acid (D) Acetophenone
- Ans. (C)
- **Sol.** When alkyl benzene is treated with acidic or alkaline KMnO<sub>4</sub>, the entire side chain is oxidized to the carboxylic acid irrespective of length of the side chain.



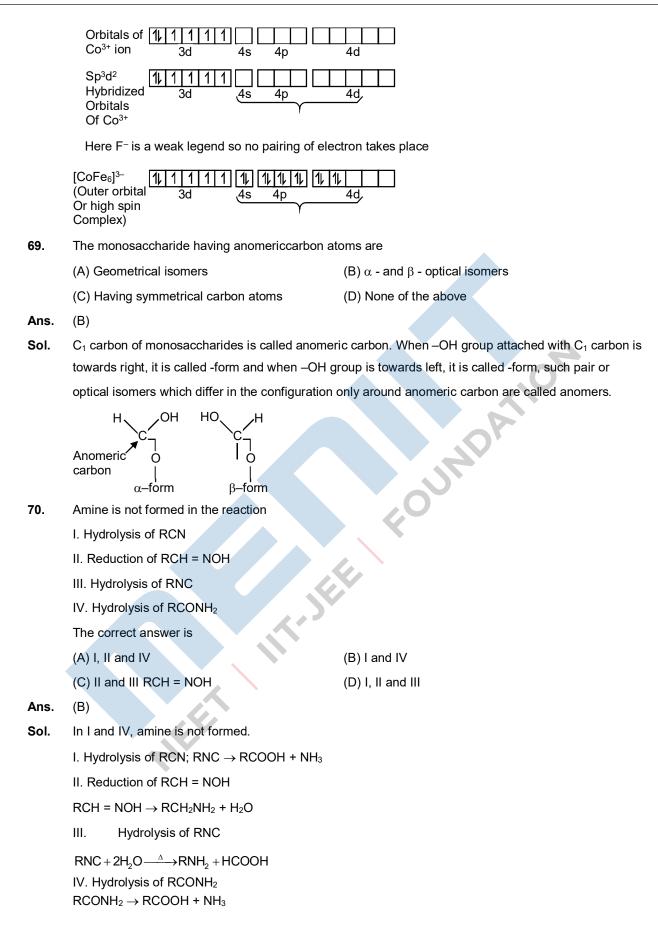
68. Which of the following is an outer d-orbital or high spin complex?

(A) [Co (NH <sub>3</sub> ) <sub>6</sub> ] <sup>3+</sup>	(B) [Ni (CN)4] <sup>2–</sup>	(C) [NiCℓ₄]²-	(D) [CoF <sub>6</sub> ] <sup>3–</sup>
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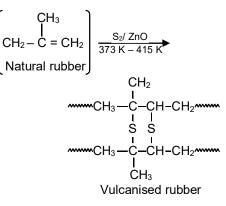
- Ans. (D)
- **Sol.**  $[Co (NH_3)_6]^{3+}$ ,  $[Ni (CN)_4]^{2-}$  and  $[NiC\ell_4]^{2-}$  are inner d-orbital or high spin complex.

[CoFe<sub>6</sub>]<sup>3–</sup>, Co = 4s<sup>2</sup>, 3d<sup>7</sup>, 4p<sup>0</sup>

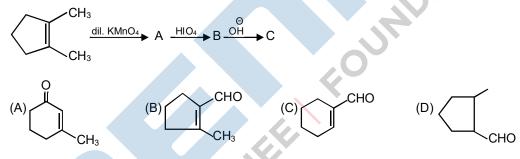
 $Co^{3+} = 3d^6, 4s^0, 4p^0$ 



- 71. In vulcanization of rubber
  - (A) Sulphur reacts to form a new compound
  - (B) Sulphur cross links are introduced
  - (C) Sulphur form a very thin protective layer on rubber
  - (D) All of the above
- Ans. (B)
- Sol. In vulcanization of rubber, sulphur cross-links are introduced at the reactive sites of double bonds.

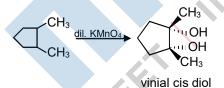


72. What will be the correct structural formula of product for the following reaction?

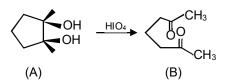


Ans. (A)

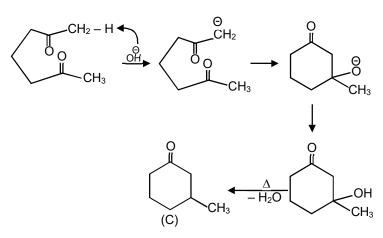
Sol. Hydroxylation reaction. The alkene on treatment with dil. KMnO<sub>4</sub> produces vicinal cis diol.



<b>Malarprade oxidation</b> cis diol undergo Malaprade oxidation in presence of HIO<sub>4</sub> and believe to proceed as



<b>Intramolecular aldol reaction</b> This diketone undergo interamolecular aldol to produce the cyclic  $\alpha$ ,  $\beta$ -unsaturated ketone as follows



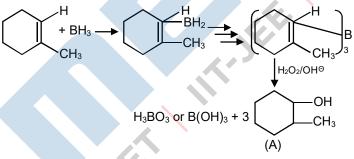
**73.** What will be the correct reaction between product when 2-methyl cyclohexane is treated with (i)  $B_2H_6$  in presence of  $H_2O_2/OH$  and (ii)  $H_2O/H_2SO_4$ 

(also consider stereochemistry of product)?

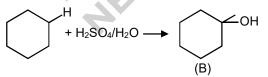
- (A) They are metamers
- (C) They are functional isomer
- (B) They are tautomers
- (D) They are positional isomer

- Ans. (D)
- **Sol.** This problem includes conceptual mixing of hydroboration. Oxidation and stereochemistry. This problem can be solved by using the skill of electrophilic addition reaction in hydroboration oxidation reaction including isomerism. The steps to solve this problems are complete reaction with acid and then identify the isomerism in the out come products.

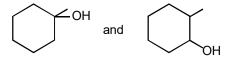
(i) <b>Hydroboration-oxidation</b> When alkene or substituted alkene is treated with  $B_2H_6$  it gives alkyl borane which on treatment with  $H_2O_2/OH^-$  causes oxidation of alkyl borone to give alcohol. The above reaction is believed to proceed as



(ii) When 2-methyl cyclohexane is treated with H<sub>2</sub>O/H<sub>2</sub>SO<sub>4</sub>



A and B have difference in position of OH only so A and B are position isomers.



74. The equilibrium constant Kp for the reaction,

$$\begin{split} N_{2}(g) + 3H_{2}(g) & \Box \ 2NH_{3} \ (g) \ is \ 1.6 \times 10^{-4} \ (atm)^{-2} \ at \ 400^{\circ}. \ What \ will \ be \ the \ equilibrium \ constant \ at \ 500^{\circ} \\ if \ heat \ of \ the \ reaction \ in \ this \ temperature \ range \ is \ -25.14 \ kCal? \\ (A) \ 1.231 \times 10^{-4} \ (atm)^{-2} & (B) \ 1.876 \times 10^{-7} \ (atm)^{-2} \\ (C) \ 1.462 \times 10^{-5} \ (atm)^{-2} & (D) \ 3.462 \times 10^{-5} \ (atm)^{-2} \\ (C) \end{split}$$

Ans. (C

**Sol.** Equilibrium constants at different temperature and heat of the reaction are related by the equation.

$$\ell n \frac{K_{p_1}}{K_{p_1}} = \frac{\Delta H^{\circ}}{R} \left[ \frac{1}{T_2} - \frac{1}{T_2} \right]$$
2.303Log  $\frac{K_{p_2}}{K_{p_1}} = \frac{\Delta H^{\circ}}{R} \left[ \frac{T_2 - T_1}{T_1 T_2} \right]$ 
Log  $K_{p_2} = \frac{-25140}{2.303 \times 2} \left[ \frac{773 - 673}{773 \times 673} \right] + \log(1.6 \times 10^{-4})$ 
log  $K_{p_2} = -4.835$ 
 $\therefore K_{p_2} = 1.462 \times 10^{-5} (atm)^{-2}$ 

**75.** At 27° C, Kp value for reaction CaCO<sub>3</sub>(s) + CaO(s) + CO<sub>2</sub>(g) is 0.1 atm The K<sub>C</sub> value for this reaction is

	(A) 4×10 <sup>−3</sup>	(B) 6 × 10 <sup>−3</sup>	(C) 2 × 10 <sup>−3</sup>	(D) 9 × 10 <sup>-3</sup>
Ans.	(A)			
Sol.	$K_p = K_c (RT)^{\Delta n}$			
	∆n = 1			
	$K_{c} = \frac{K_{p}}{RT}$ $= \frac{0.1}{0.82 \times 300}$ $= 4 \times 10^{-3}$			

76. At constant temperature and pressure which one of the following statements is correct for the reaction?

$$\begin{array}{ll} CO(g) + \frac{1}{2}O_2(g) \rightarrow CO_2(g) \\ (A) \ \Delta H = \Delta E \\ (C) \ \Delta H > \Delta E \end{array} \qquad (B) \ \Delta H < \Delta E \\ (D) \ \Delta H \ \text{is independent of physical state of reactant} \end{array}$$
Ans. (B)
Sol. As we know, \(\Delta H = \ \delta E + \ \delta RT)

where, gaseous product moles – gaseous reactant moles

For the reaction,

$$CO(g) + \frac{1}{2}O_{2}(g) \rightarrow CO(g)$$
$$\Delta n = 1 - \left(1 + \frac{1}{2}\right) = -\frac{1}{2}$$
$$\therefore \quad \Delta H = \Delta - \frac{1}{2}RT$$

Hence,  $\Delta H < \Delta E$ 

77. IUPAC name and degree of unsaturation of compound X is

- (A) 2, 3-dimethyl bicyclo [2, 2, 1] hept-5 ene, 2
- (B) 1, 2-dimethyl bicyclo [2, 2, 1] hept-4 ene, 3
- (C) 5, 6-dimethyl bicyclo [2, 2, 1] hept-2 ene, 3

(D) 4, 5-dimethyl bicycle [2, 2, 1] hept-1 ene, 2

- Ans. (C)
- **Sol.** This problem contains conceptual mixing of nomenclature of cyclic hydrocarbon and degree of unsaturation. This problem can be solved by identifying the parent chain functional group, position of functional group substituent and their position one by one and then write the name of compound according to IUPAC names then calculate degree of unsaturation.

Total carbon atom forming the bicyclic ring (hept.)

Functional group  $\Rightarrow$  double bond (ene)

Position of double bond  $\Rightarrow$  2, 2-ene

Substituents  $\Rightarrow$  2-methyl group  $\Rightarrow$  dimethyl

Position of substituents =  $5,6 \rightarrow 5,6$  -dimethyl

Number of cyclic chain =  $2 \rightarrow Bicyclo$ 

3-bridges are of 2 carbons, 2 carbons and one carbon hence,

IUPAC name = 5, 6-dimethyl bicycle [2, 2, 1] hept-2-ene

Molecular formula of compound is C9H14

Degree of unsaturation can be calculated as

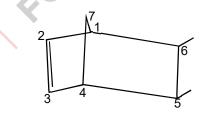
$$u = \left(C+1\right) - \frac{H}{2} + \frac{N}{2}$$

Where, u=degree of unsaturation

C = number of carbons

H = number of hydrogens

N = number of nitrogen



Х

(D) -2

Hence, for a compound having molecular formula  $C_9H_{14}$  the degree of unsaturation may be calculated as

$$u = (9-1) - \frac{14}{2} = 10 - 7 = 3$$

**78.** The oxidation state of sulphur in  $Na_2S_4O_6$  is

(A) +6 (B) 
$$+\frac{3}{2}$$
 (C)  $+\frac{5}{2}$ 

Ans. (C)

Sol. Oxidation number of Na = + 1 Oxidation number of O = -2 Let oxidation number of S = x  $\therefore$  (ON of Na) + 4(ON of O) + 6(ON of O) = 0 2(+1) + 4x + 6(-2) = 0 + 2 + 4x - 12 = 0 4x = + 12 - 2  $x = +\frac{10}{4}$  $\Rightarrow x = +\frac{5}{2}$ 

- 79. Which of the following antibiotic contains intro group attached to aromatic nucleus in its structure(A) Tetracyclin(B) Penicillin(C) Streptomycin(D)Chloramphenicol
- Ans. (C)
- **Sol.** Among the given antibiotics, only chloramphenicol contains a nitro group attached to aromatic ring. Its structure is as follows

#### Chloramphenicol

General structure of penicillin is

- 80. The behavior of the gas becomes more ideal at I. Very low pressure
  - II. Value of Z is unity
  - III. Very high pressure
  - IV. Value of Z is greater than one
  - Choose the correct option.

(A) I and II are correct (B) I and IV are correct (C) I and III are correct (D) III and IV are correct

Ans. (A)

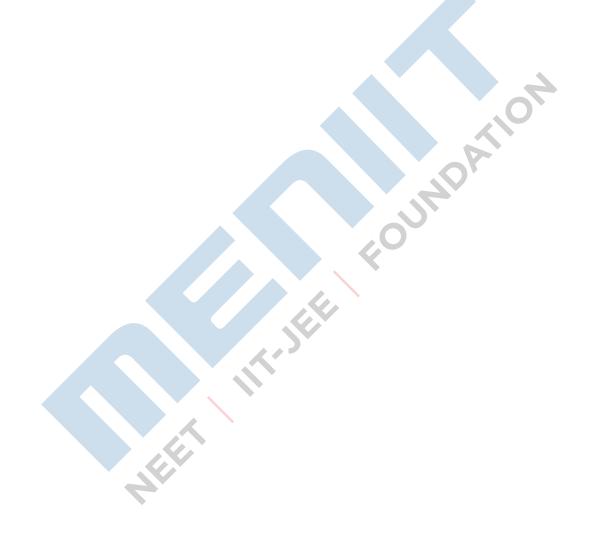
**Sol.** The deviation from ideal behavior can be measured in terms of compressibility factor Z.

 $z = \frac{pV}{nRT}$ 

For ideal gas Z = 1,

i.e., pV = nRT

At very low pressures all gases shown have  $Z \approx 1$  and behave as ideal gas. At high pressure all the gases have Z > 1 which show the deviation from ideality. At intermediate pressures, most gases have Z < 1 which also show the deviation from ideality.



### **ENGLISH**

Single correct answer type:					
1.	<b>Out of the four alte Sagacious</b>	ernatives, choose the or	ne which express the rig	ght meaning of the word.	
	(A) Shameless	(B) Wise	(C) Powerless	(D) Foolish	
Ans.	(B)				
Sol.	Sagacious means 'judi	cious', so 'wise' is correc	t answer.		
2.	<b>Out of the four alter</b>	ernatives, choose the or	ne which express the rig	ght meaning of the word.	
	Remedial				
	(A) Corrective	(B) Proficient	(C) General	(D) Optional	
Ans.	(A)				
Sol.	Remedial means 'refor	mative', so 'corrective' is	correct answer.		
				, O	
3.	<b>Out of the four alto</b>	ernatives, choose the or	ne which express the rig	ght meaning of the word.	
	Reticent				
	(A) Confident	(B) Sad	(C) Truthful	(D) Secretive	
Ans.	(D)				
Sol.	Reticent means 'quiet' so 'secretive' is correct answer.				
4.	<b>Choose the word apposite is meaning to the given word. </b>				
	Fidelity				
	(A) Faith	(B) Devotedness	(C) Allegiance	(D) Treachery	
Ans.	(D)				
Sol.	Fidelity means 'faithfulness in relations', so 'treachery' is correct antonym.				
5.	<b>Choose the word apposite is meaning to the given word. </b> Infrangible				
	(A) Complicated	(B) Breakable	(C) Weird	(D) Software	
Ans.	(B)				
Sol.	Infrangible means 'strong', so 'breakable' is correct antonym.				
	C C				
6.	<b>Choose the word a</b>	pposite is meaning to the	e given word.  Proge	eny	
	(A) Kid	(B) Parent	(C) Friend	(D) Enemy	
Ans.	(B)	、 /	· /	· · · ·	
Sol.		so 'parent' is correct ant	onvm.		
••	Progeny means 'child', so 'parent' is correct antonym.				

7.	A part of sentence is underline. Balance are given alternatives to the underlined part a, b, c and d which may improve the sentence. Choose the correct alternative.						
	It was not possible to <u>drag</u> any conclusion so he left the case.						
	(A) Fetch	(B) Find	(C) Draw	(D) No improvement			
Ans.	(C)						
Sol.	Use of 'draw' is more suitable for using before word 'conclusion', so option (draw) is correct.						
8.	A part of sentence is underline. Balance are given alternatives to the underlined part a, b, c and d which may improve the sentence. Choose the correct alternative.						
	I am <u>looking after</u> my pen which is missing.						
	(A) Looking for	(B) Looking in	(C) Looking back	(D) No improvement			
Ans.	(A)						
Sol.	Use of 'looking for' is proper because look for means 'to search for something' which suits here.						
9.	A part of sentence is underline. Balance are given alternatives to the underlined part a, b, c and d which may improve the sentence. Choose the correct alternative.						
	<u>" Mind </u> your language!" he shouted.						
	(A) Change	(B) Inspect	(C) Hold	(D) No improvement			
Ans.	(D)						
Sol.	'Mind your language' is proper to use here because it gives proper sense of sentence.						
10.	<b>Sentence Completion</b>						
		when I was student.					
	(A) Liked	(B) Used	(C) Prefer	(D) Denied			
Ans.	(B)						
Sol.	'Use to' is used when any habit is to be shown, so use of option (used) is proper.						
11.	<b>Sentence Completion</b> She was angry me.						
	(A) at	(B) about	(C) with	(D) in			
Ans.	(C)						
Sol.	(O) 'Angry' agrees with preposition 'with', so use of option (with) is correct here.						
	••••	/	,				
12.	<b>Sentence Completion</b> You should not laugh the poor.						
	(A) on	(B) at	(C) with	(D) over			
Ans.	(B)						
Sol.	Laugh agrees with preposition 'at', so use of option (at) is correct here.						

of

13.	<b>Sentence rearrangement</b>						
	1.He is a famous doctor.						
	P. Once I had to consult with him.						
	Q.I never believed him.						
	R. He suggested me a proper remedy.						
	S.I become completely fine.						
	6.Now I also admit th	is fact.					
	(A) P Q R S	(B) Q R S R	(C) Q P R S	(D) R Q S P			
Ans.	(C)						
Sol.	According to sequence of events, so option (Q P R S) is best answer.						
14.	<b>Sentence rearrangement</b>						
14.	1.We don't know the plan of Ram.						
	P. He cares for his friends.						
	Q. He is a complete person.						
	R. We want some help and advice.						
	S. As we are in a trouble.						
	6.We hope he will do his best for us.						
	(A) P R S Q	(B) Q P R S	(C) P Q R S	(D) P S R Q			
Ans.	(B)						
Sol.							
	C I						
15.	<b>Sentence rearrangement</b>						
	1.It is not my problem.						
	P. All residents of this society are careless.						
	Q.I am unable to convince anyone.						
	R. They don't want to do some good.						
	S. Every one seems to be unwise here.						
	6. We all have to suffer one day.						
	(A) P R S Q	(B) P R Q S	(C) P Q R S	(D) P S R Q			
Ans.	(A)						
Sol.							
16.	In a certain code language 'DOME' is written as '8943' and 'MEAL' is written as '4321'. What group						
10.	letters can be formed for the code '38249'?						
	(A) EOADM	(B) MEDOA	(C) EMDAO	(D) EDAMO			

Ans. (D)

Sol. As,

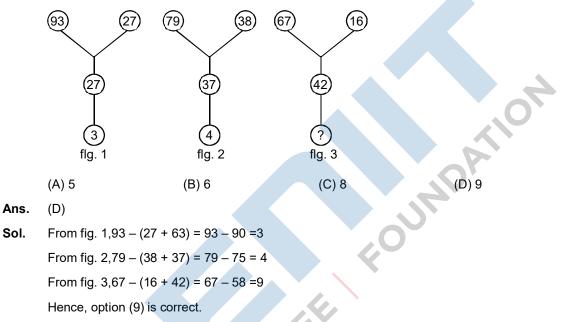
D →8	and	M →4
O <b>→</b> 9		E →3
M —→4		A →2
E —→3		L →1

In the same way 38249 will be coded as

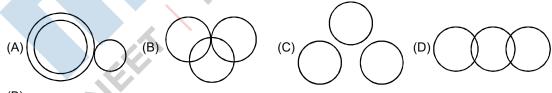
Diagram

Hence, option (EDAMO) is correct.

**17.** Find the missing number from the given response.

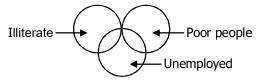


**18.** Which of the following correctly represents the relationship among illiterates, poor people and unemployed?



Ans. (B)

**Sol.** Some poor people can be unemployed, some unemployed people can be illiterates and some illiterates can be poor. Hence, correct diagram is



Hence, option (ii) is correct.

MENIIT

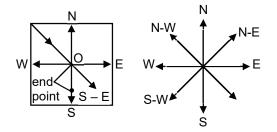
**19.** 'A' starts crossing the field diagonally from North-West. After walking half the distance, he turns right, walks some distance and turns left. Which direction is 'A' facing now?

(A) North-East (B) North-West (C) South-East

(D) South-West

Ans. (C)

Sol. Staring point



Hence, 'A' moving is South-East direction. Hence option (South-East) is correct.

**20.** In a classroom, there are 5 rows and 5 children A, B, C, D and E are seated one behind the other in 5 separate rows as follows

(B) B A C E D

- A is sitting behind C but in front of B.
- C is sitting behind E and D is sitting in front of B.
- C is sitting behind E and D is sitting in front of E.
- The order in which they are sitting from the first row to the last is

(A) D E C A B

- (C) A C B D E (D) A B E D C
- Ans. (A)
- **Sol.** From the information given in the question the arrangement of students is

 $1^{st} \mathop{\rightarrow} d$ 

 $2^{nd} \rightarrow E$ 

 $3^{rd} \to C$ 

 $4^{th} \rightarrow A$ 

 $5^{th} \rightarrow B$ 

Hence, option (D E C A B) is correct.

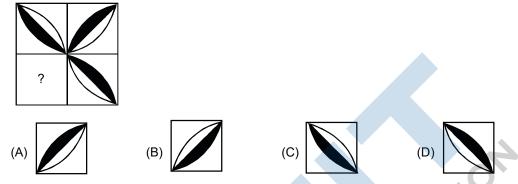
**21.** Which of the following will fill the series?

(A) 64 (B) 65 (C) 72 (D) 56

- Ans. (B)
- **Sol.** The given series follows the pattern  $1^3 + 1 = 2$

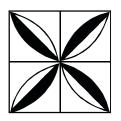
 $2^{3} + 1 = 8 + 1 = 9$   $3^{3} + 1 = 27 + 1 = 28$   $4^{3} + 1 = 64 + 1 = 65$   $5^{3} + 1 = 125 + 1 = 126$ Hence, option (65) is correct.

22. Which one of the following figures completes the original figure?



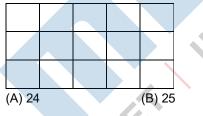
Ans. (B)

Sol. Clearly, option figure (ii) completes the original figure which looks like the figure given below



Hence, option (ii) is correct.

23. How many squares are there in the following figure?



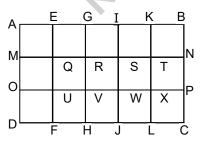
(C) 26

(D) 27

FOUN

Ans. (C)

Sol. On labeling the figure, we get



Each row contains five squares.

∴ Total number of single squares = 5 × 3 = 15

Now, combination of 4 small squares will be = 8 (i.e., AOVG, EUWI, GVXK, IWPB, MDHR, QFJS, RHLT AND SJCN)

Now, combination of 9 small squares will be = 3 (i.e., ADJI, EFLK and GHCB)

.:. Total number of squares

= 15 + 8 + 3 = 26 squares

Hence option (26) is correct.

24. Two signs in the equations have been interchanged, find out the two signs to make equation correct.

 $3 \div 5 \times 8 + 2 - 10 = 13$ 

(A) + and -

 $(C) \div and -$ 

(D) ÷ and +

Ans. (D)

Interchanging symbols + and - as given in option (+ and -) the above equation becomes Sol.

$$3 \div 5 \times 8 - 2 + 10 = \frac{3}{5} \times 8 - 2 + 10 = \frac{24}{5} + 8 \neq 13$$

MDATIC Interchanging symbols × and ÷ as given in option (× and ÷), we get

(B) × and ÷

$$3 \times \frac{5}{8} + 2 - 10 = \frac{15}{8} - 10 \neq 13$$

Interchanging symbols ÷ and – as given in option (÷ and –), we get

 $3 - 5 \times 8 + 2 \div 10$ 

$$=3+5\times\frac{8}{2}-10$$

= 3 + 20 - 10 = 13

Hence, option (+ and +) is correct.

25. <b>Assertion</b> [A] = India is a democratic country.

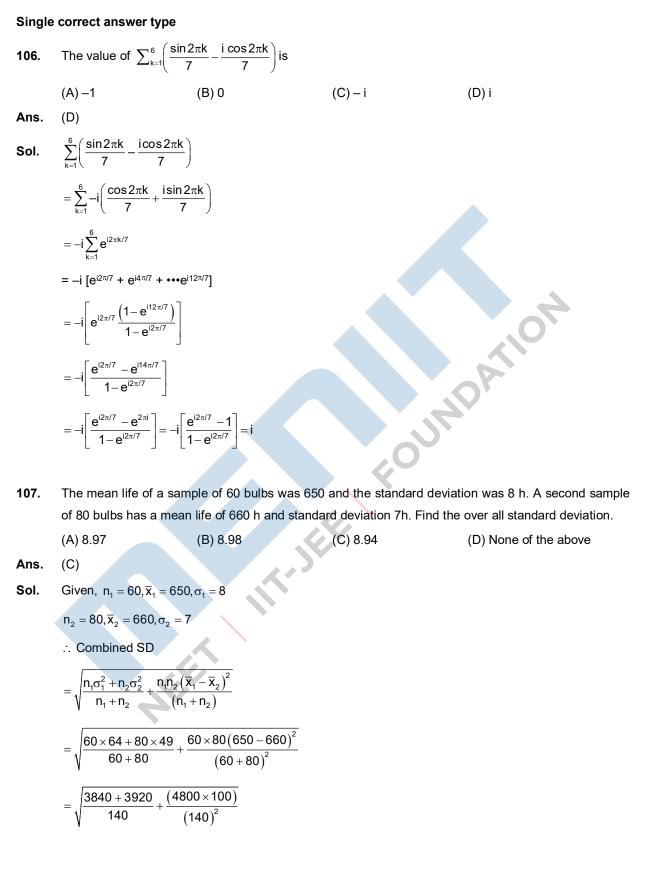
<b>Reason</b> [R] = India has a constitution of its own.

Choose the correct alternative from the given options.

- (A) Both (A) and (R) are true and (R) is correct explanation of (A)
- (B) Both (A) and (R) are true but (R) is not the correct explanation of (A)
- (C) (A) is true (R) is false
- (D) (A) is false (R) is true
- Ans. (B)
- Sol. Both Assertion and Reason are correct but India is a democratic country because the government is elected by its citizens and not because India has its own constitution.

Hence, option Both (A) and (R) are true but (R) is not the correct explanation of (A) is correct.

#### MATHEMATICS



$$= \sqrt{7760} + \frac{480000}{19600} = \sqrt{776} + \frac{4800}{196}$$

$$= \sqrt{55.42 + 24.49}$$

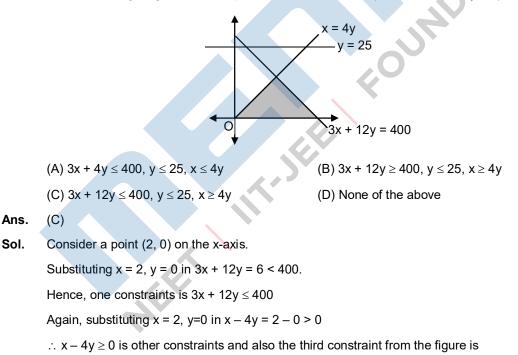
$$= \sqrt{79.91} = 8.94$$
**108.** Let R be the relation on the set R of all real numbers, defined by aRb iff  $|a - b| < 1$ , Then, R is (A) Reflexive and symmetric only (B) Reflexive and transitive only (C) Equivalence (D) None of the above
**Ans.** (A)
**Sol.** Since,  $|a - a| = 0 < 1$  so  $aRa, \forall a \in R$ 

$$\therefore \text{ R is reflexive.}$$
Now,  $aRb \Rightarrow |a - b| \le 1 \Rightarrow |b - a| \le 1 \Rightarrow bRa$ 

$$\therefore \text{ R is symmetric.}$$
But R is not transitive as
 $1R2, 2R3$  but  $1R3$  ( $\therefore |1 - 3| = 2 > 1$ )
**109.** The value of  $\int_{0}^{03} (\left[\sec^{-1}x\right] + \left[\cot^{-1}x\right]\right] dx$ , where [+] denotes the greatest integer function, is
(A)  $10\pi - \tan^{-1}x$  (B)  $8\pi - \sec 1$  (C)  $10\pi - \sec 1$  (D)  $10\pi + \sec 1$ 
**Ans.** (C)
**Sol.** Given that,
 $i = \int_{0}^{03} (\left[\sec^{-1}x\right] + \left[\cot^{-1}x\right]\right] dx + \int_{exc1}^{001} (\left[\sec^{-1}x\right] + \left[\cot^{-1}x\right] dx$ 
 $= \sup_{0}^{-1} (\left[\sec^{-1}x\right] + \left[\cot^{-1}x\right]\right] dx + \int_{exc1}^{001} (\left[\sec^{-1}x\right] + \left[\cot^{-1}x\right] dx$ 
 $= \sup_{0}^{-1} (\left[\sec^{-1}x\right] + \left[\cot^{-1}x\right]\right] dx + \int_{exc1}^{001} (\left[\sec^{-1}x\right] + \left[\cot^{-1}x\right] dx$ 
 $= \sup_{0}^{-1} (\left[\sec^{-1}x\right] + \left[\cot^{-1}x\right] dx$ 
 $= \sup_{0}^{-1} (\left[\cot^{-1}x\right] + \left[\cot^{-1}(\cos^{-1}(\tan^{-1}1))\right] \cdot is$ 
(A) 0 (B) 1 (C)  $\frac{1}{\sqrt{3}}$  (D)  $\sqrt{\frac{2}{3}}$ 
**Ans.** (D)
**Sol.**  $\sin\left[\cot^{-1}\left(\cot^{\frac{1}{3}}x\right]\right] = \sqrt{\frac{2}{3}}$ 

111. The sum of the series  $1 + 2 \cdot 2 + 3 \cdot 2^2 + 4 \cdot 2^3 + \dots + 100 \cdot 2^{99}$  is (A) 100 × 2<sup>100</sup> + 1 (D) 100 × 2<sup>100</sup>-1 (B)  $99 \times 2^{100} + 1$  (C)  $99 \times 2^{99} - 1$ Ans. (B) Sol. Let  $S = 1 + 2 \cdot 2 + 3 \cdot 2^2 + 4 \cdot 2^3 + \dots + 100 \cdot 2^{99}$ .....(i) It is an arithmetic-geometric series. On multiplying equation (i) by 2 and then subtracting, We get  $S = 1 + 2 \cdot 2 + 3 \cdot 2^2 + 4 \cdot 2^3 + \dots + 100 \cdot 2^{99}$  $2S = 1 + 2 \cdot 2 + 3 \cdot 2^2 + \dots + 99 \cdot 2^{99} + 100 \cdot 2^{100}$  $-S = 1 + 2 + 22 + 23 + \dots + 2^{99} - 100 \cdot 2^{100}$  $\Rightarrow -s = \frac{1(2^{100} - 1)}{2 - 1} - 100 \cdot 2^{100}$  $\Rightarrow -s = 2100 - 1 - 100 \cdot 2^{100}$  $\Rightarrow -S = -1 - 99 \cdot 2^{100} \Rightarrow S = 99 \cdot 2^{100} +$ 

**112.** The shaded region given below represents the constraints (other than  $x \ge 0$ ,  $y \ge 0$ )



Hence, the correct alternative is  $3x + 12y \le 400$ ,  $y \le 25$ ,  $x \ge 4y$ .

(D) 1/e

**113.** The coefficient of  $x^n$  in the expansion of  $\log_e\left(\frac{1}{1+x+x^2+x^3}\right)$ . when n is odd, is

(A) 
$$-\frac{2}{n}$$
 (B)  $-\frac{1}{n}$  (C)  $\frac{1}{n}$  (D) None of these

Ans. (B)

**Sol.**  $\log\left(\frac{1}{1+x+x^2+^3}\right) = \log\left(\frac{1-x}{1-x^4}\right)$ =  $\log(1-x) - \log(1-x^4)$ 

> When n is odd, there is no term in the second series containing x<sup>n</sup>, therefore the coefficient x<sup>n</sup> is zero in the second series and in the first series the coefficient of x<sup>n</sup> is  $-\frac{1}{n}$ . Hence, when n is odd, then, the coefficient of x<sup>n</sup> in the whole expansion is  $-\frac{1}{n} + 0 = -\frac{1}{n}$ .

> > (C) e

**114.** The maximum value of 
$$f(x) = \frac{\log x}{x}$$
 is

(A) 1 (B) 
$$\frac{2}{e}$$

Ans. (D)

$$f'(x) = \frac{1}{x^2}(1 - \log x)$$

f'(x) > 0 for x < e and f'(x) < 0 for x > e

 $\Rightarrow$  f(x) is increasing for x < e and decreasing for x > e

 $\Rightarrow$  x = e is the point of local maxima.

 $\therefore$  Maximum value of  $f(x) = \frac{1}{2}$ 

Hence, the answer is  $\frac{1}{2}$ .

**115.** Let  $\vec{a}, \vec{b}$  and  $\vec{c}$  be non-zero vectors such that no two are collinear and  $(\vec{a} \times \vec{b}) \times \vec{c} = \frac{1}{3} |\vec{b}| \vec{c} |\vec{a}|$ 

If  $\theta$  is the acute angle between the vectors  $\vec{b}$  and  $\vec{c}$  , then sin  $\theta$  is equal to

(A) 
$$\frac{2\sqrt{2}}{3}$$
 (B)  $\frac{\sqrt{2}}{3}$  (C)  $\frac{2}{3}$  (D)  $\frac{1}{2}$ 

Ans. (A)

Sol. We have, 
$$(\ddot{a} \times \ddot{b}) \times \ddot{c} = \frac{1}{3} |\ddot{b}| \vec{c} | \ddot{a}$$
  

$$\Rightarrow (\ddot{a} \cdot \ddot{c}) \ddot{b} - (\ddot{b} \cdot \ddot{c}) \ddot{a} = \frac{1}{3} |\ddot{b}| \vec{c} | \ddot{a}$$

$$\Rightarrow (\ddot{a} \cdot \ddot{c}) \ddot{b} - \left\{ (\ddot{b} \cdot \ddot{c}) + \frac{1}{3} |\ddot{b}| \vec{c} | \right\} \ddot{a} = 0$$

$$\Rightarrow (\ddot{a} \cdot \ddot{c}) = 0 \text{ and } \vec{b} \cdot \ddot{c} + \frac{1}{3} |\ddot{b}| \vec{c} | = 0$$

$$(\therefore \theta \text{ is the angle between } \ddot{b} \text{ and } \vec{c})$$

$$\Rightarrow |\ddot{b}| \vec{c}| \cos \theta + \frac{1}{3} |\ddot{b}|| \vec{c} | = 0 \Rightarrow \cos \theta = -\frac{1}{3}$$

$$\therefore \sin \theta = \sqrt{\frac{8}{9}} = \frac{2\sqrt{2}}{3}$$
116. The value of  $\lim_{x \to 0} \left(\frac{1 + 5x^2}{1 + 3x^2}\right)^{\frac{1}{x^2}}$  is  
(A)  $e^2$  (B) e (C)  $\frac{1}{e}$  (D)  $\frac{4}{e^2}$ 
Ans. (A)  
Sol.  $\lim_{x \to 0} \left(\frac{1 + 5x^2}{1 + 3x^2}\right)^{\frac{1}{x^2}} = \lim_{x \to 0} \left(1 + \frac{2x^2}{1 + 3x^2}\right)^{\frac{1}{x^2}}$ 

$$e^{\lim_{x \to 0} \left(\frac{2x^2}{1 + 3x^2}\right) = e^2$$
117. An object is observed from the points A. B and C lying in a horizontal straight line which passes

directly underneath the object. The angular elevation at B is twice that at A and at C three times that at A. If AB = a, BC = b, then the height of the object is

(A) 
$$\frac{3a}{2b}\sqrt{(a+b)(3b-a)}$$
  
(B)  $\frac{3a}{2b}\sqrt{(a+b)(3a-b)}$   
(C)  $\frac{a}{2b}\sqrt{(a+b)(3b-a)}$   
(D) None of the above  
(C)  
Let ED = h,  $\angle EAB = \alpha$ 

Ans.

Sol.

$$\therefore \angle \mathsf{EBD} = 2\alpha, \qquad \angle \mathsf{ECD} = 3\alpha$$

Now,  $\angle DBE = \angle EAB + \angle BEA$ 

$$\Rightarrow$$
  $2\alpha = \alpha + \angle BEA$ 

 $\Rightarrow$  $\angle \mathsf{BEA} = \alpha = \angle \mathsf{EAB}$ 

$$\Rightarrow AB = EB = \alpha$$
  
Similarly,  $\angle BEC = \alpha$   
From  $\triangle EBC$ ,  $\frac{BC}{\sin \alpha} = \frac{EB}{\sin(180^\circ - 3\alpha)}$   

$$\Rightarrow \frac{b}{\sin \alpha} = \frac{a}{\sin 3\alpha} \Rightarrow \frac{a}{b} = \frac{\sin 3\alpha}{\sin \alpha}$$
  

$$\Rightarrow \frac{a}{b} = \frac{3 \sin \alpha - 4 \sin^3 \alpha}{\sin \alpha} = 3 - \sin^2 \alpha$$
  

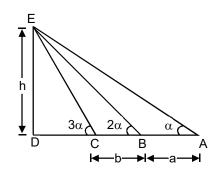
$$\Rightarrow \sin \alpha = \sqrt{\frac{3b - a}{4b}}$$
  
From  $\triangle EBD$ ,  $\sin 2\alpha = \frac{ED}{EB}$   

$$\Rightarrow ED = a \times 2 \sin \alpha \times \cos \alpha$$
  

$$\Rightarrow h = 2a\sqrt{\frac{3b - a}{4b}} \cdot \sqrt{1 - \frac{3b - a}{4b}}$$
  

$$\Rightarrow 2a\sqrt{\frac{3b - a}{4b}} \sqrt{\frac{b + a}{4b}}$$
  

$$= \frac{a}{2b}\sqrt{(a + b)(3b - a)}$$



x 4b  $= \frac{a}{2b}\sqrt{(a+b)(3b-a)}$ 118. Function f:(-∞, -1] → (0, e<sup>5</sup>] defined by f(x) = e<sup>x<sup>3</sup>-3x+2</sup> (A) many-one and onto (C) one-one and onto (D) one-one and into

.

Sol. We have, 
$$f(x) = e^{x^3-3x+2}$$
  
Let  $h(x) = x^3 - 3x + 2$   
 $\therefore h'(x) = 3x^2 - 3 = 3 (x^2 - 1)$   
 $\Rightarrow h'(x) \ge 0$  for  $x \in (-\infty, -1]$   
 $\therefore f(x)$  is increasing function.  
 $\therefore f(x)$  is one-one.  
Now, range of  $f(x) = (0, e^4]$   
But codomain of  $f(x) = (0, e^5]$   
 $\therefore f(x)$  is an into function.

**119.** The foci of the conic section  $25x^2 + 16y^2 - 150x = 175$  are(A)  $(0, \pm 3)$ (B)  $(0, \pm 2)$ (C)  $(3, \pm 3)$ (D)  $(0, \pm 1)$ 

Ans. (C) Sol. Given equation can be rewritten as  $\frac{(x-3)^2}{16} + \frac{y^2}{25} = 1$ :.  $a^2 = 16$  and  $b^2 = 25$ Now,  $e = \sqrt{1 - \frac{16}{25}} = \sqrt{\frac{25 - 16}{25}}$  $=\sqrt{\frac{9}{25}}=\frac{3}{5}$ Hence, the foci of conic section are  $(3. \pm be)$  i.e.,  $(3, \pm 3)$ . So, option  $(3, \pm 3)$  is correct 120. The system of equations x - y + 3z = 4x + z = 2 x + y - z = 0 has (A) A unique solution (B) Finitely many solution (D) None of the above (C) Infinitely many solutions FOUN Ans. (C)  $letD = \begin{vmatrix} 1 & 0 & 1 \\ 1 & 0 & 1 \\ 1 & 1 & -1 \end{vmatrix}$ Sol. = 1(10 - 1) - 1(1 + 1) + 3(1) $D_{1} = \begin{vmatrix} 4 & -1 & 3 \\ 2 & 0 & 1 \\ 0 & 1 & -1 \end{vmatrix} = 4(0-1) - 1(0+2) + 3(2)$ = -4 - 2 + 6 = 61 4 3  $D_2 = 1 \ 2 \ 1$ 1 0 –1 = 1(-1) + 4(1 + 1) + 3(0 - 2)= - 28 - 6 = 0  $D_3 = \begin{vmatrix} 1 & -1 & 4 \\ 1 & 0 & 2 \\ 1 & 1 & 0 \end{vmatrix}$ = 1(0-2) - 1(2-0) + 4(1-0)

= - 2 - 2 + 4 = 0

Hence, the given system of equations has infinitely many solutions.

121. The sum of the sequence 5,55,555, ... upto n infinite terms is

(A) 
$$\frac{5}{9} \left[ \frac{10(10^{n} - 1) + n}{9} \right]$$
  
(B)  $\frac{5}{9} \left[ \frac{10(10^{n} - 1)}{9} - n \right]$   
(C)  $\frac{5}{9} \left[ \frac{10(10^{n \pm 1} - 1)}{9} - n \right]$   
(D)  $\frac{5}{9} \left[ \frac{10(10^{n - 1} - 1)}{9} - n \right]$ 

(B) Ans.

Sol. Sn = 5 + 55 + 555 + ...upto n terms

$$= \frac{5}{9} [9 + 99 + 999....upto n \text{ temrs}]$$

$$= \frac{5}{9} [(10 - 1) + (102 - 1) + (103 - 1) + ...upto n \text{ terms}]$$

$$= \frac{5}{9} [(10 + 10^{2} + 10^{3} + ....upto n \text{ terms}) - (1 + 1 + 1 + ...upto n \text{ terms})]$$

$$= \frac{5}{9} [\frac{10(10^{n} - 1)}{10^{2}} - n]$$

.

$$9 \begin{bmatrix} 10 - 1 \\ -1 \end{bmatrix} = \frac{5}{9} \begin{bmatrix} \frac{10(10^{n} - 1)}{9} - n \end{bmatrix}$$

122. A plane passes through the point (1, -2, 3) and is parallel to the plane 2x - 2y + z = 0 distance of the point (-1, 2, 0) from the plane is

ATH

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- (D) Ans.
- Let the parallel plane to 2x 2y + z = 0 is  $2x 2y + z + \lambda = 0$ Sol. It passes through (1, -2, 3)

 $\therefore 2+4+3+\lambda=0 \Rightarrow \lambda=-9$ 

The distance of (-1, 2, 0) from the plane

$$2x - 2y + z - 9 = 0$$
 is  $\left|\frac{-2 - 4 - 9}{\sqrt{5 + 4 + 1}}\right| = \left|\frac{-15}{3}\right| = 5$ 

123. The distance between the pair of lines represented by the equation is  $x^2 - 6xy + 9y^2 + 3y - 9y - 4 = 0$  is

(A) 
$$\frac{15}{\sqrt{10}}$$
 (B)  $\frac{1}{2}$  (C)  $\sqrt{\frac{5}{2}}$  (D)  $\frac{1}{\sqrt{10}}$ 

Ans. (C)

Sol. Clearly, we have

(D) A = B ∩

(C)  $B = A \cap C$ 

FOUND

a = 1.h = -3,b = 9,g = 
$$\frac{3}{2}$$
,f =  $\frac{-9}{2}$ andc = -4  
Required distance =  $\left|2\sqrt{\frac{t^2 - bc}{b(a+b)}}\right|$ 

$$= \left| 2\sqrt{\frac{\left(\frac{-9}{2}\right)^{2} + 9 \times 4}{9(9+1)}} \right|$$
$$= \left| 2\sqrt{\frac{225}{4 \times 90}} \right| = \left| \frac{2\sqrt{5}}{2\sqrt{5}} \right| = \sqrt{\frac{5}{2}}$$

**124.** If 
$$A = \{x \in \Box : x^4 - 1 = 0\}$$

B = {x  $\in$  : x<sup>2</sup> + 1 =0}

Where  $\Box$  is complex plane.

$$(A) A = B \cup C \qquad (B) C = A \cap B$$

Ans. (A)

Sol.  $A = \{1, -1, i, -i\}$  $B = \{1, -1\}$  $C = \{i, -i\}$ 

**Now,** B C =  $\{1,-1,i,-i\}$  = A

**125.** The general solution of the differential equation 
$$\frac{dy}{dx} + \sin\left(\frac{x+y}{2}\right) = \sin\left(\frac{x-y}{2}\right)$$
  
(A)  $\log \tan\left(\frac{y}{2}\right) = C - 2\sin x$ 
(B)  $\log \tan\left(\frac{y}{4}\right) = C - 2\sin\left(\frac{x}{2}\right)$   
(C)  $\log \tan\left(\frac{y}{2} + \frac{\pi}{4}\right) = C - 2\sin x$ 
(C) None of above

Ans. (B)

Sol. We Have 
$$\frac{dy}{dx} + \sin\left(\frac{x+y}{2}\right) = \sin\left(\frac{x-y}{2}\right)$$
  
 $\Rightarrow \frac{dy}{dx} = \sin\left(\frac{x-y}{2}\right) - \sin\left(\frac{x+y}{2}\right)$   
 $\Rightarrow \frac{dy}{dx} = 2\cos\left(\frac{x}{2}\right)\sin\left(\frac{y}{2}\right)$ 

$$\Rightarrow \frac{1}{dx} = -2\cos\left(\frac{1}{2}\right)\sin\left(\frac{1}{2}\right)$$

On integrating both sides, we get

$$\begin{aligned} \int \frac{dy}{\sin\left(\frac{y}{2}\right)} &= -2\int \cos\left(\frac{x}{2}\right) dx \\ \Rightarrow \frac{1}{2}\int \cos \left(\frac{y}{2}\right) dy &= -\int \cos \left(\frac{x}{2}\right) dx \\ \Rightarrow \frac{1}{2} \cdot \left[ \frac{\log\left\{\left\{\cos \sec\left(\frac{y}{2}\right) - \cot\left(\frac{y}{2}\right)\right\}\right\}}{\frac{1}{2}} \right] &= -\frac{\sin\left(\frac{x}{2}\right)}{\frac{1}{2}} + C \\ \Rightarrow \log\left[ \frac{1}{\sin\left(\frac{y}{2}\right)} - \frac{\cos\left(\frac{y}{2}\right)}{\sin\left(\frac{y}{2}\right)} \right] &= -2\sin\left(\frac{x}{2}\right) + C \\ \Rightarrow \log\left[ \frac{2\sin^2\left(\frac{y}{4}\right)}{2\sin\left(\frac{y}{4}\right)\cos\left(\frac{y}{4}\right)} \right] &= -2\sin\left(\frac{x}{2}\right) + C \\ (\therefore 1 - \cos x = 2\sin^2\frac{x}{2} \text{ and } \sin x = 2\sin\frac{x}{2}\cos\frac{x}{2}) \\ \Rightarrow \log tan\left(\frac{y}{4}\right) &= C - 2\sin\left(\frac{x}{2}\right) \end{aligned}$$

126. The set of all real x satisfying the inequality 
$$\frac{3-|x|}{4-[x]} \ge 0$$
  
(A)  $[-3,3] \cup (-\infty,-4) \cup (4,\infty)$   
(B)  $(-\infty, -4) \cup (4,\infty)$   
(C)  $(-\infty, -3) \cup (4, \infty)$   
(D)  $(-\infty, -3) \cup (3,\infty)$   
Ans. (A)  
Sol. Given,  $\frac{3-|x|}{4-|x|} \ge 0$   
 $\Rightarrow 3 - |x| \le 0$  and  $4 - |x| < 0$   
Or  $3 - |x| \ge 0$  and  $4 - |x| < 0$   
 $\Rightarrow |x| \ge 3$  and  $|x| > 4$   
or  $|x| \le 3$  and  $|x| < 4$   
 $\Rightarrow |x| > 4$  or  $|x| \le 3$ 

$$\Rightarrow x \in (-\infty, -4) \cup [-3, 3] \cup (4, \infty)$$

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If N is the any four digit number say  $x_1$ ,  $x_2$ ,  $x_3$ ,  $x_4$  then the maximum value of  $\frac{N}{x_1 + x_2 + x_3 + x_4}$  is equal 127. to (B)  $\frac{1111}{4}$ (A) 1000 (C) 800 (D) None of these Ans. (A)  $\frac{N}{X_{1} + X_{2} + X_{3} + X_{4}}$ Sol.  $=\frac{1000x_1+100x_2+10x_3+x_4}{x_1+x_2+x_3+x_4}$  $=1000-\left(\frac{900x_2+990x_3+999x_4}{x_1+x_2+x_3+x_4}\right)$  $\Rightarrow$ Maximum value is 1000. If A and B are two events such that P(A) =0.6, P(B) = 0.2 and P $\left(\frac{A}{B}\right)$ =0.5. then P $\left(\frac{A'}{B'}\right)$ 128. equal to  $(B)^{P'(B)}$   $(J) = \frac{P(A' \cup B')}{P(B')} = \frac{P(A \cup B)'}{P(B')}$   $= \frac{1 - P(A \cup B)}{1 - P(B)}$   $\frac{1 - P(A) - P(B) + P(A \cap B)}{1 - 0.2}$   $1 - 0.6 - 0^{2}$ Ans. **Sol.**  $\therefore P(A \cap B) = P\left(\frac{A}{B}\right) P(B)$  $=\frac{1-0.6-0.2+0.1}{0.8}=\frac{3}{8}$ 129. The quartile deviation for the data is 2 3 4 5 6 x : y : 3 4 8 4 1 (B)  $\frac{1}{4}$ (C)  $\frac{1}{2}$ (A) 0 (D) 1

Ans. (D)

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Sol. Here, 
$$N = \sum f = 20$$
  
 $Q_1 = \frac{(N+1)th}{4}$  observation  
 $= \left(\frac{21}{4}\right)th$  observation  
Similarly,  $Q_2 = \frac{3(N+1)th}{4}$  observation  
 $= \left(\frac{63}{4}\right)th$  observation  
Now,  $QD = \frac{1}{2}(Q_3 - Q_1)$   
 $= \frac{1}{2}(5-3) = 1$   
130. If  $\int f(x)\cos x \, dx = \frac{1}{2}f^2(x) + C$ , then f(x) can be  
(A) x (B) 1 (C) cos x (D) sin x  
Ans. (D)  
Sol. Given that,  $\int f(x)\cos x \, dx = \frac{1}{2}f^2(x) + C$   
On differentiating w.r.t. x, we get  
 $f(x)\cos x = \frac{1}{2}.2f(x).f'(x)$   
 $\Rightarrow \cos x = f(x)$   
 $\Rightarrow \cos x = f(x)$   
 $f(x) = \sin x + C$ 

**131.** There are 10 points in a plane, out of these 6 are collinear. If ' n' is the number of triangles formed by joining these points, then

(A) 
$$n \le 100$$
 (B)  $100 < n < 140$  (C)  $140 < n \le 190$  (D)  $n > 190$ 

Ans. (A)

**Sol.** <b>Case 1</b> Taking 2 points from collinear points and one from non-collinear.

i.e., number of triangles so formed =  ${}^{6}C_{2} \times {}^{4}C_{1}$ 

$$=\frac{6\cdot 5}{1\cdot 2}\times\frac{4}{1}=60$$

<b>Case 2</b> Taking 1 point from collinear and two from non-collinear points

i.e., number of triangles so formed = ${}^{6}C_{1} \times {}^{4}C_{2}$ 

$$=\frac{6}{1}\times\frac{4\cdot3}{1\cdot2}=36$$

<b>Case 3</b> All the three points from non-collinear points.

i.e., number of triangles so formed

$$= {}^{4}C_{3} = {}^{4}C_{1}$$

$$=\frac{4}{1}=4$$

Total number of triangles = 60 + 36 + 4 = 100

<b>Alternatively</b> Number of triangles

$$=^{10} C_3 - {}^6 C_3 = \frac{10 \cdot 9 \cdot 8}{1 \cdot 2 \cdot 3} - \frac{6 \cdot 5 \cdot 4}{1 \cdot 2 \cdot 3}$$

Hence, option ( $n \le 100$ ) is correct answer.

**132.** 
$$\frac{{}^{8}C_{0}}{6} - {}^{8}C_{1} + {}^{8}C_{2} \cdot 6 - {}^{8}C_{3} - 6^{2} + {}^{8}C_{4} \cdot 6^{3} + \dots + {}^{8}C_{8} \cdot 6^{7}$$
 equals to

(B) 6<sup>7</sup>

(A) 0

(C) 6<sup>8</sup>

5<sup>8</sup> 6

Ans. (D)

**Sol.** 
$$\frac{{}^{\circ}C_0}{6} - {}^{8}C_1 + {}^{8}C_2 \cdot 6 - {}^{8}C_3 - 6^2 + {}^{8}C_4 \cdot 6^3 + \dots + {}^{8}C_8 \cdot 6^7$$

$$= \frac{1}{6} \begin{bmatrix} {}^{8}C_{0} - {}^{8}C_{1} \cdot 6^{1} + {}^{8}C_{2} \cdot 6^{2} - {}^{8}C_{3} \cdot 6^{3} + \dots + {}^{8}C_{8} \cdot 6^{8} \end{bmatrix}$$

$$=\frac{1}{6}[1-6]^{8}=\frac{1}{6}\times(-5)^{8}=\frac{5^{8}}{6}$$

Hence, the answer is  $\frac{5^8}{2}$ 

**133.** A committee of 4 students is selected at random from a group consisting 8 boys and 4 girls. Given that there is atleast one girl on the committee, then the probability that there are exactly 2 girls on the committee is

(A) 
$$\frac{68}{125}$$
 (B)  $\frac{56}{165}$  (C)  $\frac{63}{625}$  (D)  $\frac{168}{425}$ 

Ans. (D)

**Sol.** Let A denote the event that atleast one girl will be chosen and B be the event that exactly 2 girls will be chosen. We required  $P\left(\frac{B}{A}\right)$ .

Since, A denotes the vent that atleast one girl will be chosen, A' denotes that no girl is

Then, 
$$P(A') = \frac{{}^{8}C_{4}}{{}^{12}C_{4}} = \frac{70}{495} = \frac{14}{99}$$
 )

$$\Rightarrow P(A) = 1 - \frac{14}{99} = \frac{63}{99}$$

Now,  $P(A \cap B) = P$  (2 boys and 2 girls)

$$=\frac{{}^{8}C_{2} \cdot {}^{4}C_{2}}{{}^{12}C_{4}}=\frac{28 \times 6}{495}=\frac{56}{165}$$

$$\therefore P\left(\frac{B}{A}\right) = \frac{P(A \cap B)}{P(A)} = \frac{56 \times 99}{165 \times 85} = \frac{168}{425}$$

134. What are the values of for which Rolle's theorem for the function  $f(x) = x^3 - 3x^2 + 2x$  in the interval, [0, 2]- is verified?

(A) 
$$c = \pm 1$$
 (B)  $c + 1 \pm \frac{1}{\sqrt{3}}$  (C)  $c = \pm 2$  (D) None of these  
(B)  
Here, we observe that  
(a)  $f(x)$  is a polynomial, so it is continuous in the interval [0, 2].  
(b)  $f'(x) = 3x2 - 6x + 2$  exists for all  $x \in (0, 2)$   
So,  $f(x)$  is differentiable for all  $x \in (0, 2)$ 

Ans. (B)

Sol. Here, we observe that

(a) f(x) is a polynomial, so it is continuous in the interval [0, 2].

(b)  $f'(x) = 3x^2 - 6x + 2$  exists for all  $x \in (0, 2)$ 

So, f(x) is differentiable for all  $x \in (0, 2)$ 

- (C) f (0) = 0, f (2) =  $2^3 3(2)^2 + 2(2) = 0$
- ∴ f (0) = f (2)

Thus, all the three conditions of Rolle's theorem are satisfied.

So, there must exist 
$$c \in (0,2)$$
 such that f' (c) = 0

$$\Rightarrow$$
 f ' (C) = 3c<sup>2</sup> - 6c + 2 =0

$$\Rightarrow c= \Rightarrow = 1 \pm \frac{1}{\sqrt{3}} \Rightarrow C \in (0,2)$$

**135.** If 
$$\int \frac{4}{\sin^4 + \cos^4 x} dx = a \tan^{-1} \left( \frac{\tan x \frac{1}{\tan x}}{b} \right) + C$$
, then find the value of a and b, respectively.

(A) 
$$2\sqrt{2},\sqrt{2}$$
 (B)  $\sqrt{2},2$  (C)  $\sqrt{3},\sqrt{2}$  (D)  $\sqrt{2},4$ 

Ans. (A)

**Sol.** Consider that, 
$$I = \int \frac{4}{\sin^4 x + \cos^4 x} dx$$

$$I = \int \frac{4}{\cos^4 x (\tan^4 x + 1)} dx$$
$$= \int \frac{4 \sec^4 x}{1 + \tan^4 x} dx$$
$$= 4 \int \frac{\sec^4 x (1 + \tan^2 x)}{1 + \tan^4 x} dx$$

Put tan x = t  $\Rightarrow$  sec<sup>2</sup>x dx =dt

$$I = 4\int \frac{1+t^{2}}{1+t^{4}} dt = 4\int \frac{1+\frac{1}{t^{2}}}{t^{2}+\frac{1}{t^{2}}} dt$$

$$\Rightarrow I = 4\int \frac{1+\frac{1}{t^{2}}}{\left(t-\frac{1}{t}\right)+2} dt$$
Now, put  $t - \frac{1}{t} = z \Rightarrow \left(1+\frac{1}{t^{2}}\right) dt = dz$ 

$$\therefore 4\int \frac{dz}{z^{2}+\left(\sqrt{2}\right)^{2}} = \frac{4}{\sqrt{2}} \tan - 1\left(\frac{z}{\sqrt{2}}\right) + C$$

$$\Rightarrow I = 2\sqrt{2} \tan^{-1}\left(\frac{\tan x - \frac{1}{\sqrt{2}}}{\sqrt{2}}\right) + C$$

$$\therefore a = 2\sqrt{2} \ and b = \sqrt{2}$$
136. If  $A = \begin{bmatrix} -5 & -8 & 0\\ 3 & 5 & 0\\ 1 & 2 & -1 \end{bmatrix}$ , then A is  
(A) lidempotent (B) nilpotent (C) involutory (D) periodic  
Ans. (C)  
Sol. Given,  $A = \begin{bmatrix} -5 & -8 & 0\\ 3 & 5 & 0\\ 1 & 2 & -1 \end{bmatrix}$ 

$$A^{2} = \begin{bmatrix} -5 & -8 & 0\\ 3 & 5 & 0\\ 1 & 2 & -1 \end{bmatrix} \begin{bmatrix} -5 & -8 & 0\\ 3 & 5 & 0\\ 1 & 2 & -1 \end{bmatrix}$$

$$A^{2} = \begin{bmatrix} -5 & -8 & 0\\ -15 + 15 + 0 & -24 + 25 + 0 & 0 + 0 + 0\\ -5 + 6 - 1 & -8 + 10 - 2 & 0 + 0 + 1 \end{bmatrix}$$
As  $A^{2} = I$ 

**137**. The radius of the circle passing through the foci of the ellipse  $\frac{x^2}{16} + \frac{y^2}{19} = 1$  and having its centre (0,3) is

Ans. (A)

Sol.

Given, 
$$\frac{x^2}{16} + \frac{y^2}{19} = 1$$
  
=  $e = \sqrt{1 - \frac{9}{16}} = \sqrt{\frac{16 - 9}{16}} = \sqrt{\frac{7}{16}} = \frac{\sqrt{17}}{4}$ 

Coordinates of foci are  $(\pm\sqrt{7},0)$ .

Since, centre of circle is (0, 3) and passing through foci  $\left(\pm\sqrt{7},0\right)$ 

Radius of the circle 
$$=\sqrt{\left(0\pm\sqrt{7}\right)^2+\left(3-0\right)^2}$$

$$=\sqrt{7+9}=4$$

Hence, option 4 is correct.

**138**. Let L be the line of intersection of the planes 2x + 3y + z = 1 and x + 3y + 2z = 2. If L makes an angle  $\alpha$  with the positive x-axis, then  $\cos \alpha$  is equal to

(A) 
$$\frac{1}{2}$$
 (B) 1 (C)  $\frac{1}{\sqrt{2}}$  (D)  $\frac{1}{\sqrt{3}}$ 

Ans. (D)

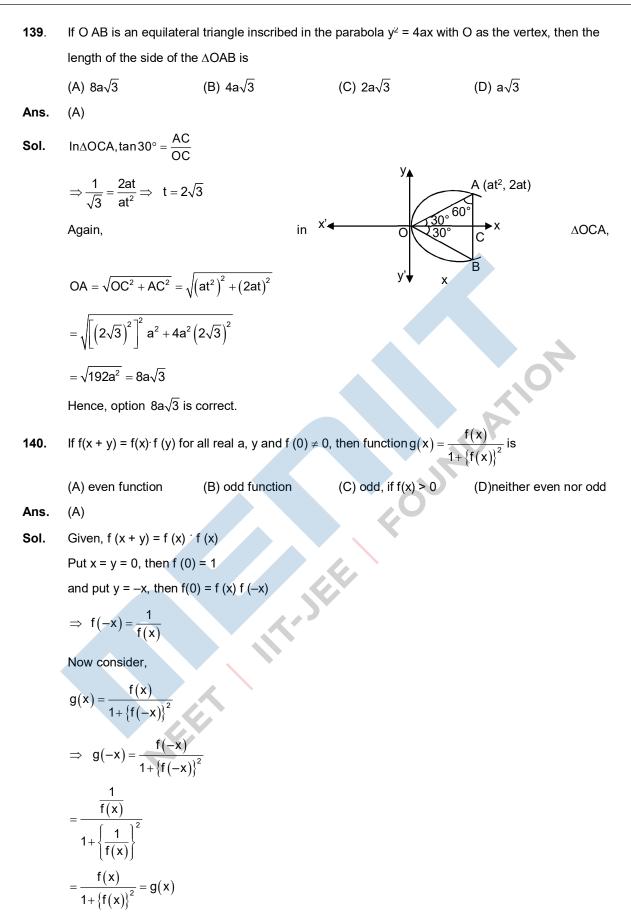
**Sol.** The two normal vectors are 
$$m = 2\hat{i} + 3\hat{j} + \hat{k}$$
 and  $n = \hat{i} + 3\hat{j} + 2\hat{k}$ 

The line L is along,  $m \times n = \begin{bmatrix} \hat{i} & \hat{j} & \hat{k} \\ 2 & 3 & 1 \\ 1 & 3 & 2 \end{bmatrix}$ 

$$=\hat{i}(6-3)-\hat{j}(4-1)+\hat{k}(6-3)$$
$$=3\hat{i}-3\hat{i}+3\hat{k}=3(\hat{i}-\hat{i}+\hat{k})$$

Now, the direction cosines of X-axis are (1, 0, 0).

$$= \cos \alpha = \frac{3(\hat{i} - \hat{j} + \hat{k}).\hat{i}}{\sqrt{3^2(1^2 + 1^2 + 1^2)\sqrt{1}}}$$
$$= \frac{3}{3\sqrt{3}} = \frac{1}{\sqrt{3}}$$
$$\Rightarrow \quad \cos \alpha = \frac{1}{\sqrt{3}}$$



141.	If $f(x) = (\tan^{-1}x)^2 + \frac{2}{\sqrt{x^2 + 1}}$ then f (x) is increasing in				
	(A) (0, ∞)	(B) (−∞ ,0)	(C) (–∞, –5)	(D) None of these	
Ans.	(A)				
Sol.	$f(x) = \left(\tan^{-1}x\right)^2 + \frac{2}{\sqrt{x^2}}$	+1			
	$f'(x) = \frac{2}{1+x^2} \left[ \tan - 1x + \frac{1}{2} + $	$-\frac{x}{\sqrt{1+x^2}}$			
	$\operatorname{Let} g(x) = \tan^{-1} - \frac{x}{\sqrt{x^2 + 1}}$				
	$\Rightarrow g'(x) = \frac{1}{1+x^2} \left[ 1 - \frac{1}{\sqrt{x^2 + 1}} \right] > 0 \text{ for all } x \in \mathbb{R}$				
	$\Rightarrow$ g(x) is increasing fo	r all $x \in R$ .			
	But $g(0) = \triangleright g(x) > 0$ for	or x > 0			
	So, $f(x) > 0$ for $x > 0$				
	Hence, f (x) is increasi	ng in (0, ∞).			
				0'	
142.	The number of solution	ns of $\cos x = [1 + \sin x], 0$	$\leq x \leq 3\pi$ is		
	(A) 1	(B) 2	(C) 3	(D) 4	
Ans.	(C)				
Sol.	Clearly, 1+ sin $x \ge 0$				
	∴ The given equation	becomes	4		
	Cos x – sin x =1				
	$\Rightarrow \cos x \cdot \frac{1}{\sqrt{2}} - \sin x \cdot \frac{1}{\sqrt{2}} = \frac{1}{\sqrt{2}}$				
	$\Rightarrow \cos\left(x + \frac{\pi}{4}\right) = \frac{1}{\sqrt{2}}$				
	$\Rightarrow X + \frac{\pi}{4} = \frac{\pi}{4}, \frac{7\pi}{4}, \frac{9\pi}{4}, \frac{15\pi}{4}$				
	$\Rightarrow x = 0, \frac{3\pi}{2}, 2\pi, \frac{7\pi}{2}$				
	$\therefore 0 \le x \le 3\pi$				
	$\Rightarrow$ x = 0, $\frac{3\pi}{2}$ , $2\pi$				

		$r GP and a^{\frac{1}{x}} = b^{\frac{1}{y}} = c^{\frac{1}{z}}, the$			
143.					
Ans.	(A) AP (A)	(B) GP	(C) HP	(D) None of these	
		1 1			
Sol.	Given, $a^{\frac{1}{x}} = b$	$\overline{y} = c^{\overline{z}}$			
	Let $a^{\frac{1}{x}} = b$	$b^{\frac{1}{y}} = c^{\frac{1}{z}} = k$			
	∴ a = k <sup>x</sup> , b = I	k <sup>y</sup> , c = k <sup>z</sup>			
	∵ a, b, c are i	n GP.			
	Therefore, b <sup>2</sup>	= ac			
	$\Rightarrow (k^{y})^{2} = k^{x}$	k <sup>2</sup>			
	$\Rightarrow k^{2\nu} = k^{x+z}$				
	Þ 2y = x + z				
	∴ x, y and z a	are in AP.			
144.	The acute ang	gle between the lines, who	se direction cosines are g	iven by	
	2ℓ – m + 2n =	0, $\ell m = mn + n\ell = 0$ , is		20.	
	(A) $\frac{\pi}{6}$	(B) $\frac{\pi}{4}$	(C) $\frac{\pi}{3}$	(D) $\frac{\pi}{2}$	
Ans.	(Α) π/6 (D)	(B) $\frac{\pi}{4}$	(C) $\frac{\pi}{3}$	(D) $\frac{\pi}{2}$	
Ans. Sol.	0		(C) $\frac{\pi}{3}$	(D) $\frac{\pi}{2}$ (i)	
	(D)	g − m + 2m =0	(C) $\frac{\pi}{3}$	L	
	(D) Given that, 2ℓ and ℓm + mn From equation	$r^{2} - m + 2m = 0$ + $n\ell = 0$		(i)	
	(D) Given that, 2ℓ and ℓm + mn From equation	$r^{2} - m + 2m = 0$ + $n\ell = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m$ + mn From equation $2\ell(\ell + n) + 2r$	$r^{2} - m + 2m = 0$ + $n\ell = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m + mn$ From equation $2\ell(\ell + n) + 2r$ $\Rightarrow 2\ell^2(\ell)$	f' - m + 2m = 0 + $n\ell = 0$ n (i), m = $2(\ell + n)$ put in equation $(\ell + n) + n\ell = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m + mn$ From equation $2\ell(\ell + n) + 2r$ $\Rightarrow 2\ell^2(\ell$ $\Rightarrow 2\ell^2 + 2\ell^2$	f' - m + 2m = 0 + $n\ell = 0$ n (i), $m = 2(\ell + n)$ put in equation n $(\ell + n) + n\ell = 0$ + n) + 2n $(\ell + n) + n\ell = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m + mn$ From equation $2\ell(\ell + n) + 2r$ $\Rightarrow 2\ell^2(\ell + n)$ $\Rightarrow 2\ell^2 + 2\ell^2$	$d^{2} - m + 2m = 0$ + $n\ell = 0$ n (i), $m = 2(\ell + n)$ put in equation n $(\ell + n) + n\ell = 0$ + $n) + 2n (\ell + n) + n\ell = 0$ $5n\ell + 2n^{2} = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m + mn$ From equation $2\ell(\ell + n) + 2r$ $\Rightarrow 2\ell^2(\ell$ $\Rightarrow 2\ell^2 + 2\ell^2$ $\Rightarrow 2\ell^2 + 2\ell^2$	$d^{2} - m + 2m = 0$ + $n\ell = 0$ n (i), $m = 2(\ell + n)$ put in equation n $(\ell + n) + n\ell = 0$ + n) + 2n $(\ell + n) + n\ell = 0$ $5n\ell + 2n^{2} = 0$ $5n\ell + 2n^{2} = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m + mn + \ell$ From equation $2\ell(\ell + n) + 2r$ $\Rightarrow 2\ell^2(\ell + n)$ $\Rightarrow 2\ell^2(\ell + n)$ $\Rightarrow 2\ell^2 + \ell$ $\Rightarrow 2\ell^2 + \ell$	$d^{2} - m + 2m = 0$ + $n\ell = 0$ n (i), $m = 2(\ell + n)$ put in equation n $(\ell + n) + n\ell = 0$ + $n) + 2n (\ell + n) + n\ell = 0$ $5n\ell + 2n^{2} = 0$ $5n\ell + 2n^{2} = 0$ $4n\ell + n\ell + 2n^{2} = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m + mn + \ell$ From equation $2\ell(\ell + n) + 2r$ $\Rightarrow 2\ell^2(\ell + \ell)$ $\Rightarrow 2\ell^2 + \ell$ $\Rightarrow \ell^2 + \ell$ $\Rightarrow \ell^2 + \ell$	f' - m + 2m = 0 + $n\ell = 0$ n (i), $m = 2(\ell + n)$ put in equation n $(\ell + n) + n\ell = 0$ + $n) + 2n (\ell + n) + n\ell = 0$ $5n\ell + 2n^2 = 0$ $5n\ell + 2n^2 = 0$ $4n\ell + n\ell + 2n^2 = 0$ - $2n) + n(\ell + 2n) = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m + mn + \ell$ From equation $2\ell(\ell + n) + 2r$ $\Rightarrow 2\ell^2(\ell + \ell)$ $\Rightarrow 2\ell^2 + \ell$ $\Rightarrow 2\ell^2 + \ell$	f' - m + 2m = 0 + $n\ell = 0$ n (i), $m = 2(\ell + n)$ put in equation n $(\ell + n) + n\ell = 0$ + $n) + 2n (\ell + n) + n\ell = 0$ $5n\ell + 2n^2 = 0$ $5n\ell + 2n^2 = 0$ $4n\ell + n\ell + 2n^2 = 0$ - $2n) + n(\ell + 2n) = 0$ (n) $(n + 2\ell) = 0$		(i)	
	(D) Given that, $2\ell$ and $\ell m + mn + \ell$ From equation $2\ell(\ell + n) + 2r$ $\Rightarrow 2\ell^2(\ell + \ell)$ $\Rightarrow 2\ell^2 + \ell$ $\Rightarrow 2\ell(\ell + \ell)$ $\ell = -2\ell$ If $\ell = -2\ell$ , then and if $n = -2\ell$	P - m + 2m = 0 + $n\ell = 0$ n (i), $m = 2(\ell + n)$ put in equation n $(\ell + n) + n\ell = 0$ + $n) + 2n(\ell + n) + n\ell = 0$ $5n\ell + 2n^2 = 0$ $5n\ell + 2n^2 = 0$ $4n\ell + n\ell + 2n^2 = 0$ $-2n) + n(\ell + 2n) = 0$ (n) $(n + 2\ell) = 0$ (2n and $n = -2\ell$		(i)	

Now, 1(-2) - (2) (-2) - 2(1) = -2 + 4 - 2 = 0

Hence, lines are perpendicular, so angle between then is  $\frac{\pi}{2}$ .

145. The equation of the lines through (1, 1) and making angles of  $45^{\circ}$  with the line x + y =0 are

(A) $x - 1 = 0$ , $x - y = 0$	(B) $x - y = 0, y - 1 = 0$
(C) x + y - 2 = 0, y - 1 = 0	(D) x − 1 = 0, y −1 = 0

(D) Ans.

Sol. Let m be the slope of required line.

$$\therefore \qquad \left| \frac{m - (-1)}{1 + m(-1)} = 1 \right|$$

$$\Rightarrow \qquad \frac{m + 1}{1 - m} = \pm 1$$

$$\Rightarrow \qquad m + 1 = 4 - m$$
and m + 1 = - 1 + m

$$\Rightarrow$$
 m = 0 and m =  $\infty$ 

: Equation of line through (1, 1) is y - 1 = 0, x - 1 = 0

Hence, option x - 1 = 0, y - 1 = 0 is correct.

The area of the figure bounded by two branches of the curve  $(y - x)^2$  and straight line x = 1 is 146.

(A) $\frac{4}{5}$ sq unit	(B) $\frac{4}{7}$ sq unit	(C) $rac{4}{9}$ sq unit	(D)
(A)			
Given curve is			
$(y - x)^2 = x^3$			

JHD

(A) Ans.

Sol. Given curve is

$$(y - x)^2 = x^3$$

$$\Rightarrow y - x = \pm x \sqrt{x}$$
  

$$\Rightarrow y = x \pm x \sqrt{x} \qquad \dots \dots (i)$$
  

$$\Rightarrow y = x - x \sqrt{x} \qquad \dots \dots (ii)$$
  
And  $x = 1$ 

From the figure, required area

$$= \int_{0}^{1} \{ (x + x\sqrt{x}) - (x - x\sqrt{x}) \} dx$$

$$= \int_{0}^{1} 2x\sqrt{x} dx = 2 \int_{0}^{1} x^{\frac{3}{2}} dx$$

$$y = x - \sqrt{x}$$

$$y = x - \sqrt{x}$$

$$y = x - \sqrt{x}$$

$$x = 1$$

$$= 2\left[\frac{x^{\frac{5}{2}}}{\frac{5}{2}}\right]_{0}^{1} = \frac{4}{5}\left[1-0\right] = \frac{4}{5}$$
 squnit

**147.** If  $(x + iy) (x + iy)^{\frac{1}{3}} = 2 + 3i$ , then 3x + 2y is equal to

(A) –20 (B) –60 (C) –120 (D) 60

Ans. (C)

**Sol.**  $(x + iy)^{\frac{1}{3}} = 2 + 3i$ 

On cubing both sides, we get

 $x + iy = (2 + 3i)^{3}$   $\Rightarrow \qquad x + iy = (2)^{3} + (3i)^{3} + 3 \times 3i (2 + 3i)$   $\Rightarrow \qquad x + iy = 8 - 27i + 18i (2 + 3i)$   $\Rightarrow \qquad x + iy = 8 - 27i + 36i - 54$   $\Rightarrow \qquad x + iy = -46 + 9i$ On comparing real and imaginary both sides,

(B) 3300

We get

X = -46, y = 9

Then, 3x + 2y = 3(-46) + 2(9)

= -138 +18 = -120

- 148. In a town of 10000 families it was found that 40% family buy newspaper A, 20% buy newspaper B and 10% families buy newspaper C, 5% families buy A and B, 3% buy B and C and 4% buy A and C. If 2% families buy all the three newspaper, then the number of families which buy A only is
  - (A) 3100

(C) 2900

ounonic

(D) 1400

Ans. (B)

**Sol.** n(A) = 40% of 10000 = 40000

n(B) = 20% of 10000 = 40000

n(C) = 10% of 10000 = 40000

n(A ∩ B) = 5% of 10000 = 500

 $n(B \cap C)$  = 3% of 10000 = 300

 $n(A \cap C) = 4\%$  of 10000 = 400

 $n(A \cap B \cap C) = 2\%$  of 10000 = 200

To find d [A  $\cap$  B<sup>c</sup>  $\cap$  C<sup>c</sup>]

 $= n[A \cap (B \cap C)^c]$ 

 $=n(A) - n[A \cap (B \cap C)]$ =n(a) - [n(A \cap B) \U n(A \cap C)] =n(A) - [n(A \cap B) + n(A \cap C) - n(A \cap B \cap C)] = 4000 - [500 + 400 - 200] = 3300

149. If 
$$|\vec{a}| = 2$$
,  $|\vec{b}| = 5$  and  $|\vec{a} \times \vec{b}| = 8$ , then  $|\vec{a} | |\vec{b}|$  is equal to  
(A) 3 (B) 4 (C) 5 (D) 6  
Ans. (D)  
Sol Since,  $\sin \theta = \frac{|\vec{a} \times \vec{b}|}{|\vec{a}||\vec{b}|} = \frac{8}{10} = \frac{4}{5}$   
 $\therefore \quad \cos \theta = \pm \frac{3}{5}$   
 $\therefore \quad \vec{a} \cdot \vec{b} = |\vec{a}| |\vec{b}| \cos \theta$   
 $= 2 \times 5 \times (\pm \frac{3}{5}) = \pm 6$   
 $\Rightarrow \quad |\vec{a} \cdot \vec{b}| = 6$ 

**150.** The equation of circle which passes through the origin and cuts off intercepts 5 and 6 from the

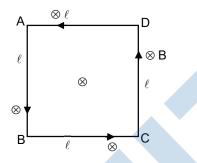
	positive parts of the x-axis and y-axis respectively is. $\left(x-\frac{5}{2}\right)^2 + \left(y-3\right)^2 = \lambda$ , where $\lambda$ is			
	(A) $\frac{61}{4}$	(B) <u>4</u>	(C) $\frac{1}{4}$	(D) 0
Ans.	(A)		K1)	
Sol.	From figure, we have			
	OP = 5, OQ = 6			
	and $OM = \frac{5}{2}, CM =$			

In  $\triangle OMC$ ,  $OC^2 = OM^2 + MC^2$ :. У▲  $OC^2 = \left(\frac{5}{2}\right)^2 + \left(3\right)^2 \Rightarrow OC = \frac{\sqrt{61}}{2}$  $\Rightarrow$ Q Thus, the required circle has its centre  $\left(\frac{5}{2},3\right)$ .  $c\left(\frac{5}{2},3\right)$ 6  $\frac{\sqrt{61}}{2}$ radius and X Ρ Hence, its equation is.  $\left(x-\frac{5}{2}\right)+\left(y-3\right)^2=\left(\frac{61}{4}\right)$ . Hence,  $\lambda = \frac{61}{4}$ FOUNDATIC AFE

# PHYSICS

#### Single correct answer type

- A square shape current loop of side length and carrying current lies in a uniform magnetic field B acting perpendicular to the plane of square loop and directed inward. The net magnetic force acting on current loop is
  - (A) IBL (B) IBL (C) zero (D) 2lbL
- Ans. (C)
- Sol. Let the current is flowing in anti-clockwise direction as shown in figure.



Now, magnetic force on AD =  $F_1$  = ilB inwards (By Fleming's left hand rule)

Similarly, magnetic force on BC =  $F_2$  = ilB inwards

Since two forces are equal in magnitude and opposite in direction, therefore, they cancel out each other.

Also, magnetic force on  $AB = F_4 = iIB$  inwards (By Fleming's left hand rule)

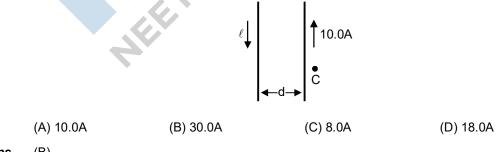
Similarly, magnetic force on  $AB = F_4 = iIB$  inwards

Again, two forces are equal in magnitude and opposite in direction, therefore, they cancel out each other.

So, the net force on the current loop is 0.

2. Two parallel conductors carry current in opposite directions, as shown in figure. One conductor carries a current of 10.0 A. Point C is a distance  $\frac{d}{2}$  to the right of the 10.0 A current. If d = 18 cm and  $\ell$  is

adjusted so that the magnetic field at C is zero, the value of the current I is



Ans. (B)

**Sol.** The magnetic field at C due to first conductor is  $B_2 = \frac{\mu_0}{2\pi} \frac{i}{\frac{3d}{2}}$  (Since Point C is separated by  $d + \frac{d}{2} = \frac{3d}{2}$ 

### from 1<sup>st</sup> conductor)

The direction of field is perpendicular to the plane of paper and directed outward. The magnetic field at

C due to second conductor is  $B_2 = \frac{\mu_0}{2\pi} \frac{i}{\frac{d}{2}}$  (Since, Point C is separated by  $\frac{d}{2}$  from 2<sup>nd</sup> conductor)

The direction of field is perpendicular to the plane of paper and directed inward.

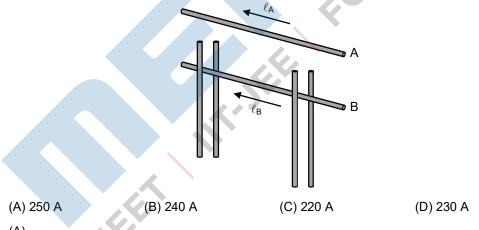
Since, direction of  $B_1$  and  $B_2$  at point C is in opposite direction and the magnetic field at C is zero, therefore,

$$B_1 = B_2$$
  
 $\mu_0 \quad i \quad \mu$ 

$$\frac{\mu_0}{2\pi}\frac{i}{\frac{3d}{2}} = \frac{\mu_0}{2\pi}\frac{10}{\frac{d}{2}}$$

On solving i = 30.0A

3. Two long, parallel conductors carry currents in the same direction, as shown in figure. Conductor A carries a current of 100A and is held firmly in position. Conductor B carries a current I<sub>B</sub> and is allowed to slide freely up and down (parallel to A) between a set of non-conducting guides. The mass per unit length of conductor B is 0.1g/cm and the distance between the two conductors is 5cm. If system of conductors is in equilibrium, the value of current I<sub>B</sub> is



Ans. (A)

Sol. When system of conductors is in equilibrium,

The magnetic force of attraction per unit length between conductors = weight of conductor B per unit length.

$$\frac{\mu_0}{2\pi} \frac{I_A \times I_B}{d} = \frac{mg}{L} = \left(\frac{m}{L}\right)g$$

$$\frac{\mu_0}{2\pi} \frac{I_A \times I_B}{d} = \left(\frac{m}{L}\right) g$$
$$2 \times 10^{-7} \times \frac{10 \times I_B}{0.05} = (0.01 \text{kg/m}) \times 10$$

On solving  $I_B$  =250 A

- 4. The number of photo electrons in a photoelectric effect experiment depends on the
  - (A) Frequency of light
  - (B) Intensity of light
  - (C) Both (frequency of light) and (intensity of light) are correct
  - (D) Both (frequency of light) and (intensity of light) are incorrect

Ans. (B)

- Sol. Number of free electrons depends on the intensity of light only.
- **5.** In hydrogen atom, if  $\lambda_1$ ,  $\lambda_2$ ,  $\lambda_3$  are shortest wavelengths in Lyman, Balmer and Paschen series respectively then  $\lambda_1$ ,  $\lambda_2$ ,  $\lambda_3$  equals
  - (A) 1 : 4 :9 (B) 9 : 4 :1 (C) 1 : 2 :3 (D) 3 : 2 :1
- Ans. (A)
- **Sol.** For hydrogen atom
  - $\frac{1}{\lambda} = R\left(\frac{1}{n_1^2} \frac{1}{n_2^2}\right), n_2 > n_1$

For Lyman  $n_1 = 1, n_2 = \infty \Rightarrow \frac{1}{\lambda_1} = R$ 

For Balmer  $n_1 = 2, n_2 = \infty \Rightarrow \frac{1}{\lambda_2} = \frac{R}{4}$ 

For Paschen series  $n_1 = 3, n_2 = \infty \Rightarrow \frac{1}{\lambda_2} = \frac{R}{9}$ 

So,  $\lambda_1 = \frac{1}{R}, \lambda_2 = \frac{4}{R}, \lambda_3 = \frac{R}{9}$  $\lambda_1; \lambda_2: \lambda_3 = 1:4:9$ 

- 6. Half-live of elements A and B are 1h and 2h respectively. Which of the following is correct?
  - (A) Element A decays slower
  - (B) Decay constant of A is smaller
  - (C) If initial number of nuclei are same then activity of A is more
  - (D) Mean life of A is more

## Ans. (C)

Sol. Let, initial number of nuclei of each element = N<sub>0</sub>

Decay constants 
$$\lambda_A = \frac{0.693}{1}hr^{-1}$$
,  $\lambda_B = \frac{0.693}{2}hr^{-2}$ 

 $\lambda_A > \lambda_B$ 

Activities  $R_A = \lambda_A N_0$ 

$$R_A = \lambda_B N_0$$

 $\Rightarrow \qquad \mathsf{R}_{\mathsf{A}} > \mathsf{R}_{\mathsf{B}} \text{ as } \lambda_{\mathsf{A}} > \lambda_{\mathsf{B}}$ 

Less half-life of element A implies faster decay

Mean life  $\tau = \frac{1}{\lambda}$   $\tau = \frac{1}{\lambda_A}, \tau_B = \frac{1}{\lambda_B}$  $\tau_A < \tau_B \text{ as } \lambda_A > \lambda_B$ 

7. A glass piece is dipped in a liquid of refractive index  $\frac{4}{3}$ , it gets disappeared in the liquid. The refractive

index of the glass piece is?

(A) 
$$\frac{4}{3}$$
 (B)  $\frac{5}{3}$  (C)  $\frac{4}{5}$  (D)  $\frac{4}{3}$ 

Ans. (D)

Sol. The glass piece will disappear only, if the refractive index of the glass and liquid is same.

So, refractive index of glass piece must be  $=\frac{1}{2}$ 

8.

If the bio-convex lens is cut as shown in the figure, the new focal length f' is

(C)  $\frac{f}{2}$ 

(A) 2f

(D) Infinite

Ans. (A)

Sol. 
$$f = (\mu - 1) \left( \frac{1}{R_1} - \frac{1}{R_2} \right)$$
$$f = (\mu - 1) \left( \frac{1}{R} + \frac{1}{R} \right)$$
$$f = (\mu - 1) \frac{2}{R}$$
.....(i)

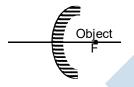
(B) f

And 
$$f' = (\mu - 1) \left( \frac{1}{R} - \frac{1}{\infty} \right)$$
  
 $f' = \frac{(\mu - 1)}{R}$  .....(ii)

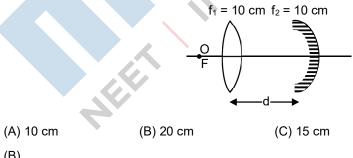
From equations (i) and (ii), we get

f' =2f

- 9. Refractive index of a medium depends
  - (A) On the medium only
  - (B) On the incident light only
  - (C) On both the conditions given in options (on the medium only) and (on the incident light only)
  - (D) None of the above
- Ans. (C)
- **Sol.** Refractive index of a medium depends on the medium as well as on the wavelength of the incident light.
- 10. A point object is placed at the focus of a convex mirror the image will be formed at(A) Infinity(B) Centre of curvature(C) At focus itself(D) None of these
- Ans. (D)
- Sol. Image will not form, because object is placed on the side from where reflection is not possible.

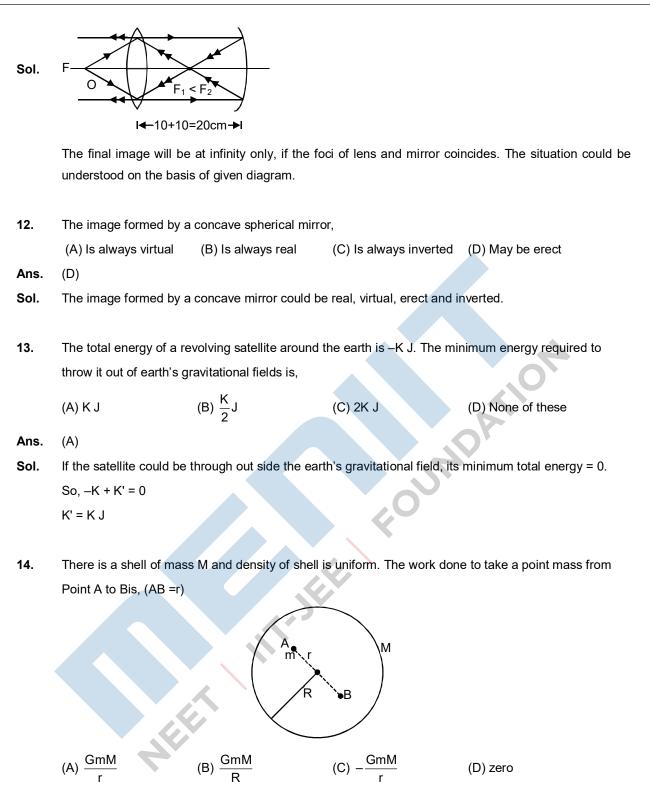


**11.** A point object is placed at the focus of the bio-convex lens. What should be the value of X, so the final image form at infinity?

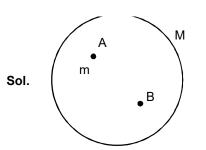


(D) None of these

Ans. (B)



Ans. (D)



The gravitational field at A and B are zero. So no work required to make charge between the points A and B.

15. A body of mass m = 20g is attached to an elastic spring of length L = 50 cm and spring constant  $k = 2 \text{ Nm}^{-1}$ . The system is revolved in a horizontal plane with a frequency u = 30 rev. Find the radius of the circular motion and the tension in the spring.

(C) 0.55 m, 0.1 N (D) 0.9 m,0.2 N (A) 0.25 (B) 0.5 m,0.52 N

- (C) Ans.
- Angular velocity  $\omega = 2\pi r = 2\pi \times \frac{30}{60} = \pi \text{ rad / s}$ Sol.

FOUNDATIC For an elastic spring force  $F = k_x$  where x is the extension.

Radius of circular motion r = L + x

Centripetal force = mr  $\omega^2$  =F

$$\Rightarrow$$
 m (L + x) $\omega^2$  = Kx

$$\Rightarrow \qquad x = \frac{mL\omega^2}{k - m\omega^2} = \frac{0.02 \times 0.5 \times (3.14)^2}{2 - 0.02 \times (3.14)^2}$$
  
= 0.05 m  
Radius of the circular motion (r)  
= L + x = 0.5 + 0.05  
= 0.55 M

≈ 0.05 m

```
Radius of the circular motion (r)
```

= L + x = 0.5 + 0.05

= 0.55 M

Tension in the spring

 $T = kx = 2 \times 0.05 \approx 0.1 N$ 

16. A gramophone record of mass M and radius R is rotating at an angular velocity w. A coin of mass m is gently placed on the record at a distance  $r = \frac{R}{2}$  from its centre. The new angular velocity of the system is (A)  $\frac{2\omega M}{(2M+m)}$  (B)  $\frac{2\omega M}{(M+2m)}$ (D)  $\frac{\omega M}{m}$ (C) ω

Ans. (A)

**Sol.** The initial angular momentum of the rotating record is  $L = I\omega$ 

Where  $I = \frac{1}{2}MR62$ 

Let  $\omega$ ' be the angular velocity of the record when the coin of mass m is placed on it at a distance r from its centre.

The angular momentum of the system becomes

Since, no external torque acts on the system, the angular momentum is conserved i.e.,

 $L' = L (I + mr^2) \omega' = I\omega$ 

or 
$$\omega' = \frac{I\omega}{I + mr^2} = \frac{\frac{1}{2}MR^2\omega}{\frac{1}{2}Mr^2 + mr^2}$$

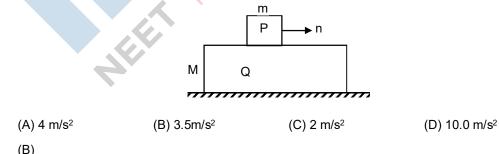
or 
$$\omega' = \frac{\omega}{1 + \frac{2mr^2}{MR^2}}$$
 .....(i)

Putting 
$$r = \frac{R}{2}$$
 in equation (i), we get

$$\omega' = \left[ \frac{\omega}{1 + \frac{2m \times \left(\frac{R}{2}\right)^2}{Mr^2}} \right]$$
$$\Rightarrow \omega' = \frac{2\omega M}{2M + m}$$

**17.** A block of mass m = 1kg is placed over a plank Q of mass M = 6 kg, placed over a smooth horizontal surface as shown in figure. Block P is given a velocity  $v = 2m/s^2$  to the right. If the coefficient of friction between P and Q is  $\mu = 0.3$ . Find the acceleration of Q relative to P.

INC



Ans. (B)

**Sol.** Frictional force between P and Q is  $f = \mu mg$  which will retard P and accelerate Q.

Retardation of P is  $a_p = -\frac{f}{m} = \frac{-\mu mg}{M} = -\mu g$ 

Acceleration of Q is  $a_{Q} = \frac{+f}{m} = \frac{\mu mg}{M}$ Acceleration Q relative to P is  $a_{QP} = a_{Q} - a_{p} = \frac{\mu mg}{M} - (-\mu g)$  $= \mu \left[ 1 + \frac{m}{M} \right]$ 

$$= 0.3 \times 10 \left[ 1 + \frac{1}{6} \right]$$

- = 3.5 m/s<sup>2</sup>
- A man runs at a speed of 4 m/s to overtake a standing bus. When he is 6 m behind the door at t = 0, the bus moves forward and continuous with a constant acceleration of 1.2 m/s<sup>2</sup> The man reaches the door in time t. Then

(C)  $4t^2 = 1.2t$ 

Μŀ

(D) 6 + 4t = 0.2t

M/B

В

4t

(A)  $4t = 6 + 0.6t^2$  (B)  $1.2t^2 = 4t$ 

Ans. (A)

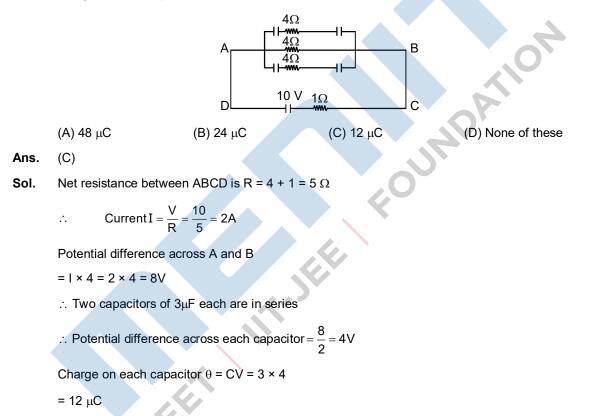
Sol. Let us draw the figure for given situation,

$$\Rightarrow \qquad 4t = 6 + \frac{1}{2} \times 1.2 \times t^2$$
$$\Rightarrow \qquad 4t = 6 + 0.6 t^2$$

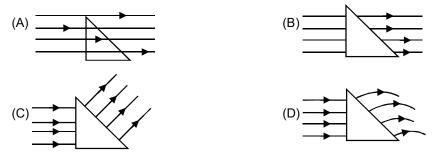
- **19.** In completely inelastic collision
  - (A) The complete KE of the medium must lost
  - (B) The linear momentum of the system must remain conserved during collision
  - (C) Both (the complete KE of the medium must lost) and (the linear momentum of the system must remain conserved during collision) are correct
  - (D) Both (the complete KE of the medium must lost) and (the linear momentum of the system must remain conserved during collision) are incorrect
- Ans. (B)
- Sol. In any type of collision the linear moments of the system remain conserved even during collision.
- **20.** The number of particles pen unit volume is given by  $n = -\frac{Dn_2 n_1}{x_2 x_1}$  are crossing a unit area perpendicular to x-axis in unit time, when n<sub>1</sub> and n<sub>2</sub> are the number of particles per unit volume for the values x<sub>1</sub> and x<sub>2</sub> of x respectively. Then the dimensional formula of diffusion constant D is (A) [M<sup>0</sup>LT<sup>0</sup>] (B) [M<sup>0</sup>L<sup>2</sup>T<sup>-4</sup>] (C) [M<sup>0</sup>LT<sup>-3</sup>] (D) [M<sup>0</sup>L<sup>2</sup>T<sup>-1</sup>]

Ans. (D)

- Sol. From the given relation,  $D = \frac{n(x_2 x_1)}{n_2 n_1}$  $Here, [n] = \left[\frac{1}{area \times time}\right] = \frac{1}{\left[L^2 T\right]} \left[L 2T^{-1}\right]$  $x_2 x_1 = [L] \text{ and } n_2 n_1 = \left[\frac{1}{volume}\right] = \left[\frac{1}{L^3}\right]$  $= [L^{-3}]$ So,  $[D] = \frac{\left[L^{-2} T^{-1} L\right]}{\left[L^{-3}\right]} = \left[L^2 T^{-1}\right]$
- **21.** In the given circuit (as shown in figure). Each capacitor has a capacity of 3μF. What will be the net charge on each capacitor?



**22.** A solid conductor is placed in an uniform electric field as shown in figure. Which path will the lines of force follow?



Ans. (C)

- Sol. The electric field inside a conductor is zero and is always perpendicular to the surface of a conductor.
- **23.** A bomb at rest explodes into three parts of the same mass. The linear momentum of two parts are  $-2P\hat{i}$  and  $P\hat{j}$  The magnitude of momentum of third part is  $P\sqrt{x}$ . Find.

(A) P (B) √5P (C)2P (D) 10P

Ans. (B)

**Sol.** Given,  $P_1 = -2P\hat{i} = 2P$  along negative x-axis.

 $P_2 = P\hat{i} = P$  along y-axis.

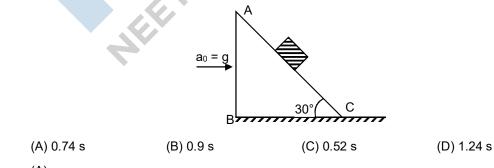
The resultant momentum of two parts

$$P'\sqrt{P_1^2 + P_2^2} = \sqrt{(2P)^2} = P\sqrt{5}$$

As the bomb was initially at rest final momentum of all the three parts must be zero  $P_3 + P' = 0$ 

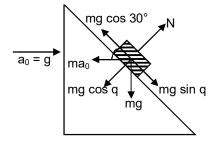
Block is placed on an inclined plane. The block is moving towards right horizontally with an acceleration a<sub>0</sub> = g. The length of the inclined plane (AC) is equal to 1m. Whole the situation are shown in the figure. Assume that all the surfaces are frictionless. The time taken by the block to reach

From C to A is (take  $g = 10 \text{ m/s}^2$ )



Ans. (A)

Sol. The forces on smaller block is given as



For the motion of the block along the incline plane in upward direction.

Net force on the block = mass × acceleration of the block

 $\Rightarrow \qquad \mbox{mg cos } 30^\circ - \mbox{mg sin mg cos } 30^\circ - \mbox{mg sin } 0^\circ = \mbox{ma} \qquad (\because a_0 = g)$ 

$$a = \left(\frac{\sqrt{3}-1}{2}\right)g = 3.66 \,\text{m/s}^2$$

Now, from equation of motion  $s = \frac{1}{2}at^2$ 

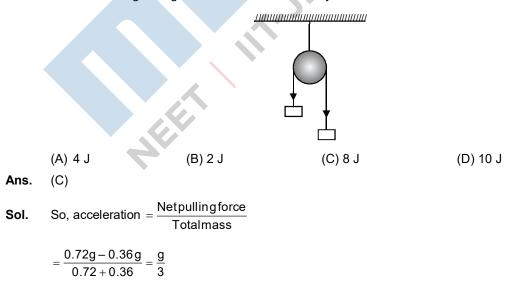
$$\Rightarrow \qquad t = \sqrt{\frac{2s}{a}} = \sqrt{\frac{2 \times 1}{3.66}} = 0.74 \, s$$

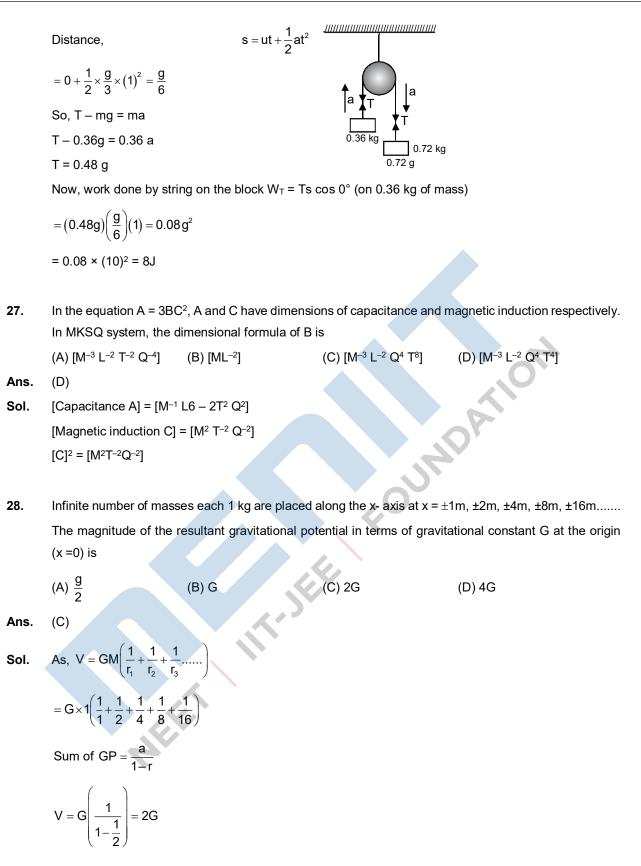
- 25. Pseudo force is
  - (A) Electromagnetic in nature
  - (C) A gravitational force

- (B) A nuclear force
- (D) None of the above

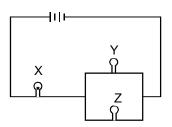
Ans. (D)

- Sol. Pseudo force is not a real force.
- **26.** A light in extensible string that goes over a smooth fixed pulley as shown in the figure connect two blocks of masses 0.36kg and 0.72kg. Taking  $g = 10 \text{ n/s}^2$ . Find the work done by string on the block of mass 0.36 kg during the first second after the system is released from rest.





**29.** Three bulbs X, Y and Z are connected as shown in figure. The bulbs Y and Z are identical. If bulb gets fused then,



- (A) Both X and Y will glow more brightly
- (B) Both X and Y will glow less brightly
- (C) X will glow less brightly and Y will glow more brightly
- (D) X will glow more brightly and Y will glow less brightly
- Ans. (C)
- **Sol.** If bulb Z is fused, the current stops flowing through Z. The effective resistance of the circuit due to bulbs X and Y in series becomes more as compared to before. Due to which, the current in the circuit

JUND

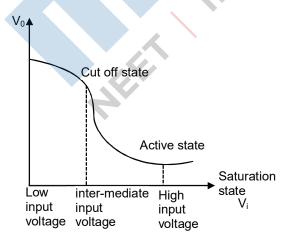
decrease.

 $\therefore$  brightness  $\propto$  (current)<sup>2</sup>

So, the brightness of bulb × decreases,

Now, bulb Y gets more current than before fusing the bulb, Z.

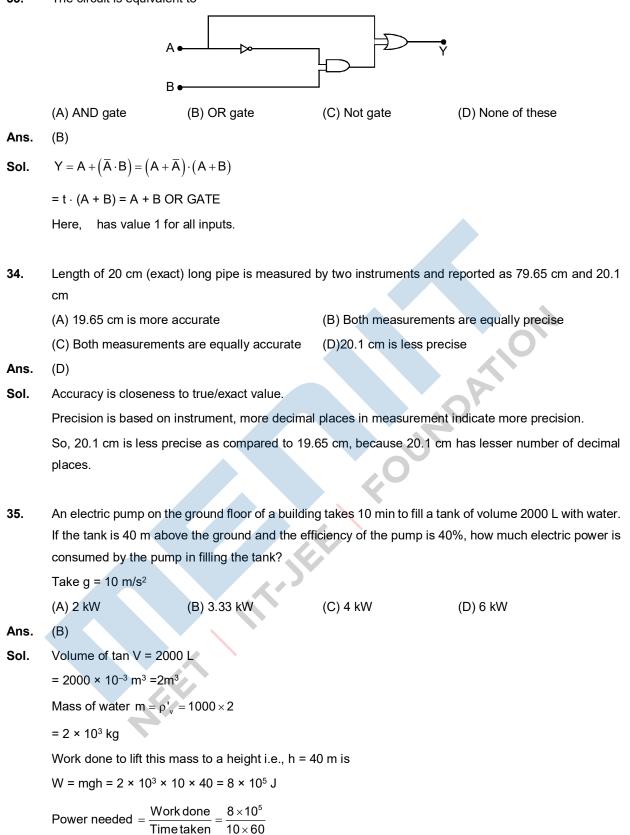
- ... Brightness of bulb Y will increase.
- 30. Active state of n-p-n transistor, in circuit is achieved by
  - (A) Low input voltage
  - (B) High input voltage
  - (C) Both (low input voltage) and (high input voltage)
  - (D) Neither (low input voltage) nor (high input voltage)
- Ans. (D)
- Sol. Transfer characteristic of transistor is



Active state is achieved at intermediate input voltage.

31. A turntable of radius R = 10 m is rotation making 98 rev in 10 s with a boy of mass m = 60 kg standing at its centre. He starts running along a radius. Find the frequency of the turntable when the boy is 4m from the centre. The moment of inertia of the turntable about its axis 1000 kg - m<sup>2</sup>. (A) 10 Hz (B) 2.5 Hz (C) 5 Hz (D) 4 Hz (C) Ans. Sol. Initial moment of inertia of the system is  $M_1$  = moment of inertia of turntable + Moment of inertia of boy at the centre = 1000 + 0 = 1000 kgm<sup>2</sup> Initial frequency v = 9.8 rev/sFinal moment of the system  $M_2$  = MI of turntable + MI of boy at a distance 4m from the centre of turntable.  $= 1000 + 60 \times (4)^2$ = 1960 kgm<sup>2</sup> Since no external torque acts, the angular momentum of the system is conserved i.e.  $I_2\omega_2 = I_1\omega_1$  $I_2v_2 = I_1v_1$ →  $v_2 = \frac{I_1 v_1}{I_2} = \frac{1000 \times 9.8}{1960}$ = 5 rev/s = 5Hz 32. To transmit a signal, if height of transmitting signal above surface of the earth is H, this signal can be received on surface of the earth upto distance d from transmitter. Then (C) d ∝ H<sup>±</sup>́ (D) d  $\propto H^{\frac{3}{2}}$ (B) d  $\propto$  H<sup>2</sup> (A)  $D \propto H$ Ans. (C)  $OB^2 + BC^2 = OC^2$ Sol. С  $R^2 + BC^2 = (R + H)^2$  $BC\sqrt{2RH+H^2}$  $=\sqrt{RH}\left(2\frac{H}{R}\right)$ R R Here,  $\frac{H}{R} \ll 2$ Centre of Earth So, BC =  $\sqrt{2RH}$  $BC \approx AB = d = distance of reach$  $d = \sqrt{2RH} \propto H^{\frac{1}{2}}$ R = constant = 6400 km = radius of the earth





$$P' = \frac{4}{3} \times 10^{3} W$$
It P is the total power consumed, the useful power available = 40%, if P' = 0.4 P, then  

$$= 0.4P = \frac{4}{3} \times 10^{3}$$

$$\Rightarrow P = 3.33 \times 10^{3} W$$
36. A vessel containing 1 mole of O<sub>2</sub> gas (molar mass 32) at temperature T. The pressure of the gas is p.  
An identical vessel containing one mole of He gas (molar mass 4) at temperature. 2T has a pressure of  
(A)  $\frac{P}{8}$  (B) P (C) 2p (D) 8p  
Ans. (C)  
Sol. Applying gas equation pV = nRT  
We can write:  $p_{1V} = n_{1}RT_{1}$   
 $\Rightarrow P_{2} = \frac{n_{2}}{n_{1}} \times \frac{T_{2}}{T_{1}}$   
 $= \frac{1}{p_{1}} \times \frac{T_{1}}{n_{1}} \times \frac{T_{2}}{T_{1}}$   
 $= \frac{1}{p_{1}} \times \frac{T_{1}}{n_{1}} \times \frac{T_{2}}{T_{1}}$   
(A) 37% (B) 11% (C) 33% (D) 15.5%  
Ans. (D)  
Sol. We know  $v_{em} = \sqrt{\frac{3RT}{M}}$   
 $\Rightarrow v_{increaseinv_{em}} = \frac{\sqrt{\frac{3RT}{M}} \cdot \frac{\sqrt{3RT_{1}}}{\sqrt{\frac{3RT_{1}}{M}}} \times 100$   
 $= \frac{\sqrt{400} - \sqrt{300}}{\sqrt{300}} \times 100$   
 $= \frac{20 - 17.32}{17.32} \times 100$   
 $= 15.5\%$ 

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38. A particle of mass m = 5g is executing simple harmonic motion with an amplitude 0.3 m and time period

 $\frac{\pi}{5}$  sec. The maximum value of force acting on the particle is

(A) 5 N (C) 0.5 N (B) 4 N (D) 0.15 N

Ans. (D)

Sol. We know,

Maximum acceleration  $a_{max} = \omega^2 A \frac{4\pi^2}{T^2} A$ 

$$=\frac{4\pi^2}{\left(\frac{\pi}{5}\right)^2}\times 0.3=30\,\text{m/s}^2$$

Maximum force

$$F_{max} = ma_{max} = \frac{5}{1000} \times 30 = 0.15 N$$

39. A partition wall has two layers of different materials A and B in contact with each other. They have the same thickness but the thermal conductivity of layer A is twice that of B. At steady state if the temperature difference across the layer B is 50 K, then the corresponding difference across the layer A is

(C) 25 K

(D) 60 K

Ans. (C)

Sol. Let T be the junction temperature Тв Here,  $K_A = 2K_B$ ,  $T - T_B = 50K$ At the steady state  $H_A = H_B$  $\frac{K_{A}A(T_{A}-T)}{L} = \frac{K_{B}A(T-T_{B})}{L}$ В А  $\Rightarrow$  $2K_{B}(T_{A}-T) = K_{B}(T-T_{B})$  $\Rightarrow$  $T_A - T = \frac{T - T_B}{2} = \frac{50}{2} = 25K$  $\Rightarrow$ Pulse rate of a normal person is 75 per min. The time period of heart is 40. (A) 0.08 s (B) 0.75 s (C) 1.25 s (D) 1.75 s Ans. (A) Sol. The beat frequency of heart is

(B) 12.5 K

$$v = \frac{75}{(1 \text{ min})} = \frac{75}{60 \text{ s}} = 1.25 \text{ s}^{-1}$$
  
= 1.25 Hz  
The time period of heart is  
$$T = \frac{1}{v} = \frac{1}{1.25 \text{ s}^{-1}} = 0.8 \text{ s}$$